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## LEVY 09/626,026

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L56 ANSWER 1 OF 2 HCAPLUS COPYRIGHT 2002 ACS 2002:107417 HCAPLUS ACCESSION NUMBER:

DOCUMENT NUMBER: 136:151974

TITLE:

Polyurethane dispersions in alcohol-water

system with good stability

Kantner, Steven S.; Scholz, Matthew T.; INVENTOR(S):

Lewandowski, Kevin M.

PATENT ASSIGNEE(S): 3M Innovative Properties Company, USA

PCT Int. Appl., 47 pp. SOURCE:

CODEN: PIXXD2

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1

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                                                    APPLICATION NO.
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                                                  WO 2001-US19186 20010614
                                  20020207
     WO 2002010243
                          A1
          W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN,
                CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH,
          CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, F1, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, TZ, UA, UG, UZ, VN, YU, ZA, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM

RW: GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZW, AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG
PRIORITY APPLN. INFO.:
                                                  US 2000-626028
                                                                      A 20000727
      A polyurethane dispersion stable in a mixt. of alc. and water is
      useful in cold-seal adhesives, cosmetic applications, medical goods, etc.
     The dispersion comprises a reaction product of (a) .qtoreq.1
      isocyanate-terminated polyurethane prepolymer derived from (i)
      .gtoreq.1 oligomeric polyactive hydrogen compd. (insol. in the alc.) which
      is an alkyl, aryl, or aralkyl structure, optionally substituted by N, O,
      and S; (ii) .gtoreq.1 polyisocyanate, and (iii) .gtoreq.1 polyactive
      hydrogen compd. sol. in the mixt. of alc. and water; (b) a polyfunctional
     chain extender; and (c) a monofunctional chain terminator; wherein the equiv. ratio of difunctional chain extender to prepolymer isocyanate is
      0.60-1.20. Thus, 60 parts urethane prepolymer prepd. from isophorone
      diisocyanate, Kraton L 2203 (hydrogenated polybutadiene diol); Terathane
      2000 (polytetramethylene oxide diol), sodiosulfopolyester diol (di-Me
      5-sodiosulfoisophthalate-diethylene glycol-.epsilon.-caprolactone
      copolymer) and Surfynol 104 surfactant was mixed with 1.48 parts
      ethylenediamine (chain extender) in 76.5 parts ethanol and 13.5 parts
      water, coated onto a polyester film and dried at 70.degree. For 10 min,
      showing adhesion peel from glass 2 N/dM and peel from self 88 N/dM, resp.
      9003-17-2P
IT
      RL: BUU (Biological use, unclassified); IMF (Industrial manufacture); POF
      (Polymer in formulation); TEM (Technical or engineered material use); BIOL
      (Biological study); PREP (Preparation); USES (Uses)
          (butadiene rubber, hydroxy-terminated, polymers with polyols,
         polyisocyanates, diamines and hydroxy-contg. amines;
         polyurethane dispersions in alc.-water system with good
         stability)
      9003-17-2 HCAPLUS
RN
      1,3-Butadiene, homopolymer (9CI) (CA INDEX NAME)
CN
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### LEVY 09/626,026

CRN 106-99-0 CMF C4 H6

 $H_2C = CH - CH = CH_2$ TΤ 107-15-3DP, Ethylene diamine, polymers with polyurethanes 124-22-1DP, Dodecylamine, reaction products with polyurethanes 124-30-1DP, Octadecylamine, reaction products with polyurethanes 124-68-5DP, 2-Amino-2-methyl-1-propanol, polymers with polyurethanes 3007-31-6DP, Didodecylamine, reaction products with polyurethanes 3965-55-7DP, Dimethyl5sodiosulfoisophthalate, polymers with polyols, polyisocyanates, diamines and hydroxy-contg. amines 4098-71-9DP, IPDI, polymers with polyols, diamines and hydroxy-contg. amines 25190-06-1DP, Terathane 2000, polymers with polyols, polyisocyanates, diamines and hydroxy-contg. amines 25322-68-3DP, Polyethylene glycol, polymers with polyols, polyisocyanates, diamines and hydroxy-contg. amines 25322-69-4DP, Polypropylene glycol, polymers with polyols, polyisocyanates, diamines and hydroxy-contg. amines 394728-16-6P 394728-18-8P 394728-19-9P 394728-21-3P **394728-23-5DP**, reaction products with amines **394728-24-6DP** , reaction products with amines 394728-25-7P 394728-27-9DP, reaction products with amines 394728-29-1P 394728-31-5P 394728-33-7P 394730-29-1P 395068-49-2P RL: BUU (Biological use, unclassified); IMF (Industrial manufacture); POF (Polymer in formulation); TEM (Technical or engineered material use); BIOL (Biological study); PREP (Preparation); USES (Uses) (polyurethane dispersions in alc.-water system with good stability) RN 107-15-3 HCAPLUS CN 1,2-Ethanediamine (9CI) (CA INDEX NAME) H2N-CH2-CH2-NH2 124-22-1 HCAPLUS RN1-Dodecanamine (9CI) (CA INDEX NAME) CN  $H_2N-(CH_2)_{11}-Me$ RN 124-30-1 HCAPLUS 1-Octadecanamine (9CI) (CA INDEX NAME) CN  $H_2N-(CH_2)_{17}-Me$ 124-68-5 HCAPLUS RN 1-Propanol, 2-amino-2-methyl- (8CI, 9CI) (CA INDEX NAME) CN

RN 3007-31-6 HCAPLUS

CN 1-Dodecanamine, N-dodecyl- (9CI) (CA INDEX NAME)

 $Me^- (CH_2)_{11} - NH^- (CH_2)_{11} - Me$ 

RN 3965-55-7 HCAPLUS

CN 1,3-Benzenedicarboxylic acid, 5-sulfo-, 1,3-dimethyl ester, sodium salt (9CI) (CA INDEX NAME)

## Na

RN 4098-71-9 HCAPLUS

CN Cyclohexane, 5-isocyanato-1-(isocyanatomethyl)-1,3,3-trimethyl- (9CI) (CA INDEX NAME)

RN 25190-06-1 HCAPLUS

CN Poly(oxy-1,4-butanediyl), .alpha.-hydro-.omega.-hydroxy- (9CI) (CA INDEX NAME)

RN 25322-68-3 HCAPLUS

CN Poly(oxy-1,2-ethanediyl), .alpha.-hydro-.omega.-hydroxy- (9CI) (CA INDEX

NAME)

$$HO = \begin{bmatrix} CH_2 - CH_2 - O \end{bmatrix}_n$$

25322-69-4 HCAPLUS RN

CN Poly[oxy(methyl-1,2-ethanediyl)], .alpha.-hydro-.omega.-hydroxy- (9CI) (CA INDEX NAME)

$$HO \longrightarrow (C_3H_6) - O \longrightarrow n$$

394728-16-6 HCAPLUS RN

CN 1,3-Benzenedicarboxylic acid, 5-sulfo-, 1,3-dimethyl ester, sodium salt, polymer with 1,2-ethanediamine, .alpha.-hydro-.omega.-hydroxypoly(oxy-1,4butanediyl), 5-isocyanato-1-(isocyanatomethyl)-1,3,3-trimethylcyclohexane, Kraton Liquid L 2203, 2-oxepanone, 2,2'-oxybis[ethanol] and 2,4,7,9-tetramethyl-5-decyne-4,7-diol, block (9CI) (CA INDEX NAME)

CM 1

CRN 191617-94-4

CMF Unspecified

CCI PMS, MAN

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

CM

CRN 25190-06-1

CMF (C4 H8 O)n H2 O

CCI PMS

HO 
$$\left[ (CH_2)_4 - O \right]_n$$

CM 3

4098-71-9 CRN

C12 H18 N2 O2 CMF

CRN 3965-55-7 CMF C10 H10 O7 S . Na

Na

CM 5

CRN 502-44-3 CMF C6 H10 O2

CM 6

CRN 126-86-3 CMF C14 H26 O2

CM 7

CRN 111-46-6 CMF C4 H10 O3

 ${\tt HO-CH_2-CH_2-O-CH_2-CH_2-OH}$ 

```
8
     CM
     CRN 107-15-3
     CMF C2 H8 N2
H_2N-CH_2-CH_2-NH_2
RN
     394728-18-8 HCAPLUS
CN
    1,3-Benzenedicarboxylic acid, 5-sulfo-, 1,3-dimethyl ester, sodium salt,
    polymer with 2-amino-2-methyl-1-propanol, 1,2-ethanediamine,
     .alpha.-hydro-.omega.-hydroxypoly(oxy-1,4-butanediyl),
     .alpha.-hydro-.omega.-hydroxypoly(oxy-1,2-ethanediyl),
     .alpha.-hydro-.omega.-hydroxypoly[oxy(methyl-1,2-ethanediyl)],
     5-isocyanato-1-(isocyanatomethyl)-1,3,3-trimethylcyclohexane and Kraton
    Liquid L 2203, block (9CI) (CA INDEX NAME)
     CM
          1
    CRN
         191617-94-4
    CMF
          Unspecified
    CCI
          PMS, MAN
*** STRUCTURE DIAGRAM IS NOT AVAILABLE ***
    CM
          2
    CRN
          25322-69-4
         (C3 H6 O)n H2 O
    CMF
    CCI
         IDS, PMS
    CDES 8:ID
       - (C3H6) - O-
    CM
          3
    CRN
          25322-68-3
    CMF
          (C2 H4 O)n H2 O
    CCI
          PMS
        СН2-СН2-О
    CM
```

25190-06-1

PMS

(C4 H8 O)n H2 O

CRN

CMF CCI

$$\begin{array}{c|c} \text{HO} & \hline & \text{(CH2)} \, 4 - \text{O} \\ \hline & n \end{array}$$

CRN 4098-71-9 CMF C12 H18 N2 O2

CM 6

CRN 3965-55-7 CMF C10 H10 O7 S . Na

Na

CM 7

CRN 124-68-5 CMF C4 H11 N O

CM 8

CRN 107-15-3 CMF C2 H8 N2

H2N-CH2-CH2-NH2

RN 394728-19-9 HCAPLUS

CN 1,3-Benzenedicarboxylic acid, 5-sulfo-, 1,3-dimethyl ester, sodium salt, polymer with 2-amino-2-methyl-1-propanol, 1,2-ethanediamine, .alpha.-hydro-.omega.-hydroxypoly(oxy-1,2-ethanediyl), .alpha.-hydro-.omega.-hydroxypoly[oxy(methyl-1,2-ethanediyl)], 5-isocyanato-1-(isocyanatomethyl)-1,3,3-trimethylcyclohexane and Kraton Liquid L 2203, block (9CI) (CA INDEX NAME)

CM 1

CRN 191617-94-4 CMF Unspecified CCI PMS, MAN

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

CM 2

CRN 25322-69-4 CMF (C3 H6 O)n H2 O CCI IDS, PMS CDES 8:ID

$$HO = \begin{bmatrix} (C_3H_6) - O \end{bmatrix}_n H$$

CM 3

CRN 25322-68-3 CMF (C2 H4 O)n H2 O CCI PMS

$$\begin{array}{c|c} \text{HO} & \hline & \text{CH}_2\text{--}\text{CH}_2\text{--}\text{O} & \hline & \text{H} \end{array}$$

CM 4

CRN 4098-71-9 CMF C12 H18 N2 O2

CRN 3965-55-7 CMF C10 H10 O7 S . Na

Na

CM 6

CRN 124-68-5 CMF C4 H11 N O

CM 7

CRN 107-15-3 CMF C2 H8 N2

 $H_2N-CH_2-CH_2-NH_2$ 

RN 394728-21-3 HCAPLUS

CN 1,3-Benzenedicarboxylic acid, 5-sulfo-, 1,3-dimethyl ester, sodium salt, polymer with 1,2-ethanediamine, 1,2-ethanediol, .alpha.-hydro-.omega.-hydroxypoly(oxy-1,2-ethanediyl), 5-isocyanato-1-(isocyanatomethyl)-1,3,3-trimethylcyclohexane, Kraton Liquid L 2203 and 1,2-propanediol, block

(9CI) (CA INDEX NAME)

CM 1

CRN 191617-94-4 CMF Unspecified CCI PMS, MAN

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

CM 2

CRN 25322-68-3 CMF (C2 H4 O)n H2 O CCI PMS

$$HO = \begin{bmatrix} CH_2 - CH_2 - O \end{bmatrix}_n H$$

CM 3

CRN 4098-71-9 CMF C12 H18 N2 O2

CM 4

CRN 3965-55-7 CMF C10 H10 O7 S . Na

Na

CRN 107-15-3 CMF C2 H8 N2

 $H_2N-CH_2-CH_2-NH_2$ 

CM 6

CRN 75-56-9 CMF C3 H6 O



CM 7

CRN 75-21-8 CMF C2 H4 O



RN 394728-23-5 HCAPLUS

CN 1,3-Benzenedicarboxylic acid, 5-sulfo-, 1,3-dimethyl ester, sodium salt, polymer with 1,2-ethanediamine, .alpha.-hydro-.omega.-hydroxypoly(oxy-1,4-butanediyl), .alpha.-hydro-.omega.-hydroxypoly(oxy-1,2-ethanediyl), .alpha.-hydro-.omega.-hydroxypoly[oxy(methyl-1,2-ethanediyl)], 5-isocyanato-1-(isocyanatomethyl)-1,3,3-trimethylcyclohexane and Kraton Liquid L 2203, block (9CI) (CA INDEX NAME)

CM 1 .

CRN 191617-94-4 CMF Unspecified CCI PMS, MAN

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

CM 2

CRN 25322-69-4 CMF (C3 H6 O)n H2 O CCI IDS, PMS CDES 8:ID

$$HO \longrightarrow (C_3H_6) - O \longrightarrow H$$

CRN 25322-68-3 CMF (C2 H4 O)n H2 O

CCI PMS

$$HO - \begin{bmatrix} CH_2 - CH_2 - O \end{bmatrix}_n H$$

CM 4

CRN 25190-06-1

CMF (C4 H8 O)n H2 O

CCI PMS

$$HO - \left[ -(CH_2)_4 - O \right]_n H$$

CM 5

CRN 4098-71-9 CMF C12 H18 N2 O2

CM 6

CRN 3965-55-7

CMF C10 H10 O7 S . Na

Na

CM7

CRN 107-15-3 CMF C2 H8 N2

 $H_2N-CH_2-CH_2-NH_2$ 

RN 394728-24-6 HCAPLUS

CN 1,3-Benzenedicarboxylic acid, 5-sulfo-, 1,3-dimethyl ester, sodium salt, polymer with 1,2-ethanediamine, .alpha.-hydro-.omega.-hydroxypoly(oxy-1,4butanediyl), .alpha.-hydro-.omega.-hydroxypoly(oxy-1,2-ethanediyl), .alpha.-hydro-.omega.-hydroxypoly[oxy(methyl-1,2-ethanediyl)], 5-isocyanato-1-(isocyanatomethyl)-1,3,3-trimethylcyclohexane, Pripol 1009 and Pripol 2033, block (9CI) (CA INDEX NAME)

CM 1

158516-85-9 CRN Unspecified CMF MAN

CCI

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

CM2

CRN 127290-22-6 Unspecified CMF

CCI MAN

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

CM 3

CRN 25322-69-4 CMF (C3 H6, O)n H2 O CCI IDS, PMS CDES 8:ID

$$HO = \begin{bmatrix} (C3H6) - O \end{bmatrix} \frac{1}{n} H$$

CRN 25322-68-3 CMF (C2 H4 O)n H2 O CCI PMS

$$HO = \begin{bmatrix} CH_2 - CH_2 - O \end{bmatrix}_n H$$

CM 5

CRN 25190-06-1 CMF (C4 H8 O)n H2 O CCI PMS

$$HO \longrightarrow \begin{bmatrix} & & \\ & & \\ & & \end{bmatrix} \begin{pmatrix} CH_2 \end{pmatrix} \begin{pmatrix} & & \\ & & \\ & & \end{pmatrix} \begin{pmatrix} & & \\ &$$

CM 6

CRN 4098-71-9 CMF C12 H18 N2 O2

CM .

CRN 3965-55-7 CMF C10 H10 O7 S . Na

CCI IDS, PMS CDES 8:ID

$$HO - \left[ -(C_3H_6) - O - \right]_n H$$

CRN 25322-68-3 · CMF (C2 H4 O)n H2 O CCI PMS

CM 5

CRN 25190-06-1 CMF (C4 H8 O)n H2 O CCI PMS

HO 
$$\left[ (CH_2)_4 - O \right]_n$$
 H

CM 6

CRN 4098-71-9 CMF C12 H18 N2 O2

CM T

CRN 3965-55-7 CMF C10 H10 O7 S . Na

Na

CM 8

CRN 124-68-5 CMF C4 H11 N O

$$\begin{array}{c} \text{NH}_2 \\ | \\ \text{Me-C-CH}_2 - \text{OH} \\ | \\ \text{Me} \end{array}$$

CM 9

CRN 107-15-3 CMF C2 H8 N2

H2N-CH2-CH2-NH2

RN 394728-27-9 HCAPLUS

CN 1,3-Benzenedicarboxylic acid, 5-sulfo-, 1,3-dimethyl ester, sodium salt, polymer with 1,2-ethanediamine, .alpha.-hydro-.omega.-hydroxypoly(oxy-1,2-ethanediyl), .alpha.-hydro-.omega.-hydroxypoly[oxy(methyl-1,2-ethanediyl)], 5-isocyanato-1-(isocyanatomethyl)-1,3,3-trimethylcyclohexane and Kraton Liquid L 2203, block (9CI) (CA INDEX NAME)

CM :

CRN 191617-94-4 CMF Unspecified CCI PMS, MAN

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

CM 2

CRN 25322-69-4 CMF (C3 H6 O)n H2 O CCI IDS, PMS CDES 8:ID

$$HO = \begin{bmatrix} (C3H_6) - O \end{bmatrix}_n H$$

CRN 25322-68-3 CMF (C2 H4 O)n H2 O CCI PMS

$$HO - CH_2 - CH_2 - O - I$$

CM 4

CRN 4098-71-9 CMF C12 H18 N2 O2

CM 5

CRN 3965-55-7 CMF C10 H10 O7 S . Na

Na

CM 6

CRN 107-15-3 CMF C2 H8 N2

 $H_2N-CH_2-CH_2-NH_2$ 

RN 394728-29-1 HCAPLUS

CN Propanoic acid, 3-hydroxy-2-(hydroxymethyl)-2-methyl-, polymer with 2-amino-2-methyl-1-propanol, 1,2-ethanediamine, .alpha.-hydro-.omega.-hydroxypoly(oxy-1,4-butanediyl), 5-isocyanato-1-(isocyanatomethyl)-1,3,3-trimethylcyclohexane and Kraton Liquid L 2203 (9CI) (CA INDEX NAME)

CM 1

CRN 191617-94-4 CMF Unspecified CCI PMS, MAN

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

CM 2

CRN 25190-06-1 CMF (C4 H8 O)n H2 O CCI PMS

CM 3

CRN 4767-03-7 CMF C5 H10 O4

$$\begin{array}{c} & \text{Me} \\ | \\ \text{HO-CH}_2\text{--C-CO}_2\text{H} \\ | \\ \text{CH}_2\text{--OH} \end{array}$$

CM 4

CRN 4098-71-9 CMF C12 H18 N2 O2

CRN 25190-06-1 CMF (C4 H8 O)n H2 O CCI PMS

HO-
$$\left(\text{CH}_2\right)_4$$
-O- $\left(\text{D}_1\right)_n$ 

CM 4

CRN 4098-71-9 CMF C12 H18 N2 O2

CM 5

CRN 124-68-5 CMF C4 H11 N O

CM 6

CRN 107-15-3 CMF C2 H8 N2

 $H_2N-CH_2-CH_2-NH_2$ 

CRN 191617-94-4 CMF Unspecified CCI PMS, MAN

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

CM 2

CRN 25190-06-1 CMF (C4 H8 O)n H2 O

CCI PMS

CM 3

CRN 4098-71-9 CMF C12 H18 N2 O2

CM 4

CRN 124-68-5 CMF C4 H11 N O

CM 5

CRN 107-15-3 CMF C2 H8 N2

 $H_2N-CH_2-CH_2-NH_2$ 

RN 394730-29-1 HCAPLUS

CN 1,3-Benzenedicarboxylic acid, 5-sulfo-, 1,3-dimethyl ester, sodium salt, polymer with .alpha.-hydro-.omega.-hydroxypoly(oxy-1,2-ethanediyl), 5-isocyanato-1-(isocyanatomethyl)-1,3,3-trimethylcyclohexane, Kraton Liquid L 2203, methyloxirane, methyloxirane polymer with oxirane bis(2-aminopropyl) ether and oxirane, block (9CI) (CA INDEX NAME)

CM 1

CRN 191617-94-4 CMF Unspecified CCI PMS, MAN

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

CM 2

CRN 25322-68-3 CMF (C2 H4 O)n H2 O CCI PMS

$$HO \longrightarrow CH_2 - CH_2 - O \longrightarrow n$$

CM 3

CRN 4098-71-9 CMF C12 H18 N2 O2

CM 4

CRN 3965-55-7 CMF C10 H10 O7 S . Na

Na

CM 5

CRN 75-56-9 CMF C3 H6 O

CM

CRN 75-21-8 CMF C2 H4 O



7 CM

CRN 65605-36-9

CMF C3 H9 N O . 1/2 (C3 H6 O . C2 H4 O) x CDES 8:GD

> CM8

> > CRN 6168-72-5 CMF C3 H9 N O

$$^{
m NH_2}_{
m H_3C-CH-CH_2-OH}$$

CM

CRN 9003-11-6

```
CMF (C3 H6 O . C2 H4 O)x
CCI PMS

CM 10

CRN 75-56-9
CMF C3 H6 O
```



CRN 75-21-8 CMF C2 H4 O



```
RN
     395068-49-2 HCAPLUS
     1,3-Benzenedicarboxylic acid, 5-sulfo-, 1,3-dimethyl ester, sodium salt,
CN
    polymer with 2-amino-2-methyl-1-propanol, 1,2-ethanediamine,
     .alpha.-hydro-.omega.-hydroxypoly(oxy-1,2-ethanediyl),
     .alpha.-hydro-.omega.-hydroxypoly[oxy(methyl-1,2-ethanediyl)],
     5-isocyanato-1-(isocyanatomethyl)-1,3,3-trimethylcyclohexane, Kraton
    Liquid L 2203, methyloxirane and oxirane, block (9CI) (CA INDEX NAME)
     CM
          1
    CRN
         191617-94-4
         Unspecified
    CMF
    CCI
         PMS, MAN
*** STRUCTURE DIAGRAM IS NOT AVAILABLE ***
          2
    CM
    CRN
         25322-69-4
    CMF
          (C3 H6 O)n H2 O
     CCI
         IDS, PMS
    CDES 8:ID
```

$$HO = \begin{bmatrix} (C_3H_6) - O \end{bmatrix}_n H$$

CM 3

CRN 25322-68-3 CMF (C2 H4 O)n H2 O CCI PMS

$$HO = \begin{bmatrix} CH_2 - CH_2 - O \end{bmatrix}_n H$$

CRN 4098-71-9 CMF C12 H18 N2 O2

CM 5

CRN 3965-55-7 CMF C10 H10 O7 S . Na

Na

CM 6

CRN 124-68-5 CMF C4 H11 N O

CM 7

# LEVY 09/626,026

CRN 107-15-3 CMF C2 H8 N2

 $H_2N-CH_2-CH_2-NH_2$ 

CM 8

CRN 75-56-9 CMF C3 H6 O



CM 9

CRN 75-21-8 CMF C2 H4 O



18 183274-48-8P, Dimethyl 5-sodiosulfoisophthalate-polyethylene
glycol copolymer 220118-45-6P 394728-13-3P
RL: IMF (Industrial manufacture); RCT (Reactant); PREP (Preparation); RACT
(Reactant or reagent)

(polyurethane dispersions in alc.-water system with good stability)

RN 183274-48-8 HCAPLUS

CN 1,3-Benzenedicarboxylic acid, sulfo-, 1,3-dimethyl ester, sodium salt, polymer with .alpha.-hydro-.omega.-hydroxypoly(oxy-1,2-ethanediyl) (9CI) (CA INDEX NAME)

CM 1

CRN 30675-44-6

CMF C10 H10 O7 S . Na

CCI IDS

CDES 8:ID, RING

D1-SO3H

Na

CM 2

CRN 25322-68-3

CMF (C2 H4 O)n H2 O

CCI PMS

$$HO = \begin{bmatrix} CH_2 - CH_2 - O \end{bmatrix}_n I$$

RN 220118-45-6 HCAPLUS

CN 1,3-Benzenedicarboxylic acid, 5-sulfo-, 1,3-dimethyl ester, sodium salt, polymer with .alpha.-hydro-.omega.-hydroxypoly(oxy-1,2-ethanediyl) and .alpha.-hydro-.omega.-hydroxypoly[oxy(methyl-1,2-ethanediyl)], block (9CI) (CA INDEX NAME)

CM 1

CRN 25322-69-4 CMF (C3 H6 O)n H2 O

CCI IDS, PMS

CDES 8:ID

CM 2

CRN 25322-68-3

CMF (C2 H4 O)n H2 O

CCI PMS

$$HO = \begin{bmatrix} CH_2 - CH_2 - O \end{bmatrix}_n H$$

CRN 3965-55-7 CMF C10 H10 O7 S . Na

## Na

RN 394728-13-3 HCAPLUS

CN 1,3-Benzenedicarboxylic acid, 5-sulfo-, 1,3-dimethyl ester, sodium salt, polymer with 2-oxepanone and 2,2'-oxybis[ethanol], block (9CI) (CA INDEX NAME)

CM 1

CRN 3965-55-7 CMF C10 H10 O7 S . Na

Na

CM 2

CRN 502-44-3 CMF C6 H10 O2

CRN 111-46-6 CMF C4 H10 O3

HO-CH2-CH2-O-CH2-CH2-OH

RN 64-17-5 HCAPLUS

CN Ethanol (9CI) (CA INDEX NAME)

H<sub>3</sub>C-СH<sub>2</sub>-ОН

RN 67-63-0 HCAPLUS CN 2-Propanol (9CI) (CA INDEX NAME)

ОН | Н3С-СН-СН3

RN 71-23-8 HCAPLUS CN 1-Propanol (9CI) (CA INDEX NAME)

H<sub>3</sub>C-СH<sub>2</sub>-СH<sub>2</sub>-ОН

RN 7732-18-5 HCAPLUS CN Water (8CI, 9CI) (CA INDEX NAME)

H20

REFERENCE COUNT: 4 THERE ARE 4 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

### LEVY 09/626,026

### => d ibib abs hitstr 2

L56 ANSWER 2 OF 2 HCAPLUS COPYRIGHT 2002 ACS ACCESSION NUMBER: 2002:107416 HCAPLUS

DOCUMENT NUMBER: 136:167821

TITLE: Polyurethane film-forming dispersions in

alcohol-water system and their manufacture for use in

antimicrobial compositions

INVENTOR(S): Scholz, Matthew T.; Kantner, Steven S.;

Comstock, Kristen L.; Brown, Christopher J.

PATENT ASSIGNEE(S): 3M Innovative Properties Company, USA

SOURCE: PCT Int. Appl., 68 pp.

CODEN: PIXXD2

DOCUMENT TYPE:

LANGUAGE:

Patent English

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

```
PATENT NO.
                      KIND DATE
                                            APPLICATION NO.
                                                              DATE
                             20020207
     WO 2002010242
                       Α1
                                            WO 2000-US32962 20001204
         W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN,
             CR, CU, CZ, DE, DK, DM, DZ, EE, ES, FI, GB, GD, GE, GH, GM, HR,
             HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT,
             LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, PL, PT, RO, RU,
             SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, TZ, UA, UG, UZ, VN, YU,
             ZA, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM
         RW: GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZW, AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, TR, BF,
             BJ, CF, CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG
                                                          A 20000727
PRIORITY APPLN. INFO.:
                                          US 2000-627110
     The dispersion is stable in an lower alc.-H2O mixt. The dispersion is (a)
     prepolymer reaction product of (i) .gtoreq.1 oligomeric polyactive H
     compd., where the compd. is an alkyl, aryl, or aralkyl structure
     optionally substituted in and/or on the chain by N, O, S and combinations,
     and where the compd. is insol. in 50:50 (wt%) alc.-H2O mixt., (ii)
     .gtoreq.1 polyisocyanate, and (iii) .gtoreq.1 polyactive H compd. sol. in
     the alc.-H2O mixt. selected from compd. contg. an ionic group, a compd.
     contg. a moiety capable of forming an ionic group, a compd. contg. a
     polyester, polyether, or polycarbonate group having a C/O ratio .ltoreq.5,
     and mixts., and (b) .gtoreq.1 polyfunctional chain extender. Thus, an
     example polymer was formed from Kraton L 2203, Terathane 2000,
     sodiosulfopolyester diol, Surfynol 102 surfactant, Desmodur I, and
     ethylenediamine chain extender.
IT
     25214-14-6DP, Adipic acid-1,6-hexanediol-neopentyl glycol
     copolymer, polyurethane derivs.
```

RL: IMF (Industrial manufacture); POF (Polymer in formulation); THU (Therapeutic use); BIOL (Biological study); PREP (Preparation); USES (Uses)

(Fomrez E 65-56; **polyurethane** dispersions in iso-Pr alc.-water system and manuf. for use in antimicrobial compns. and film-forming surgical drapes)

RN 25214-14-6 HCAPLUS

CN Hexanedioic acid, polymer with 2,2-dimethyl-1,3-propanediol and 1,6-hexanediol (9CI) (CA INDEX NAME)

CM 1

CRN 629-11-8

CMF C6 H14 O2

 ${\rm HO^-}$  (CH<sub>2</sub>)<sub>6</sub> $-{\rm OH}$ 

CM 2

CRN 126-30-7 CMF C5 H12 O2

CM 3

CRN 124-04-9 CMF C6 H10 O4

 $HO_2C - (CH_2)_4 - CO_2H$ 

CN Iodine (8CI, 9CI) (CA INDEX NAME)

I-I

RN 7681-55-2 HCAPLUS CN Iodic acid (HIO3), sodium salt (8CI, 9CI) (CA INDEX NAME)

Na

RN 7758-05-6 HCAPLUS
CN Iodic acid (HIO3), potassium salt (8CI, 9CI) (CA INDEX NAME)

K

RN 18472-51-0 HCAPLUS

CN D-Gluconic acid, compd. with N,N''-bis(4-chlorophenyl)-3,12-diimino-2,4,11,13-tetraazatetradecanediimidamide (2:1) (9CI) (CA INDEX NAME)

CM 1

CRN 526-95-4 CMF C6 H12 O7 CDES 5:D-GLUCO

Absolute stereochemistry.

CM 2

CRN 55-56-1 CMF C22 H30 C12 N10

RN 25655-41-8 HCAPLUS

CN 2-Pyrrolidinone, 1-ethenyl-, homopolymer, compd. with iodine (9CI) (CA INDEX NAME)

CM 1

CRN 7553-56-2

CMF I2

I-I

CM 2

CRN 9003-39-8

RN 7491-02-3 HCAPLUS CN Decanedioic acid, bis(1-methylethyl) ester (9CI) (CA INDEX NAME)

CN Cellulose, 2-hydroxyethyl 2-[2-hydroxy-3-(trimethylammonio)propoxy]ethyl 2-hydroxy-3-(trimethylammonio)propyl ether, chloride (9CI) (CA INDEX NAME)

CM 1

CRN 170553-71-6

CMF C8 H20 N O3 . x C6 H16 N O2 . x C2 H6 O2 . x Unspecified CDES 8:GD

CM 2

CRN 170344-46-4 CMF C8 H20 N O3

```
OH
Me_3+N-CH_2-CH-CH_2-O-CH_2-CH_2-OH
          CM
               3
          CRN 44814-66-6
          CMF C6 H16 N O2
        OH
HO-CH_2-CH-CH_2-N+Me_3
          CM
          CRN
               9004-34-6
          CMF
               Unspecified
               PMS, MAN
          CCI
*** STRUCTURE DIAGRAM IS NOT AVAILABLE ***
          CM
               5
          CRN 107-21-1 .
CMF C2 H6 O2
HO-CH_2-CH_2-OH
ΙT
     96-27-5DP, 3-Mercapto-1,2-propanediol, polyurethane
     derivs. 107-15-3DP, Ethylenediamine, polyurethane
     derivs. 79103-62-1DP, Desmodur W, polyurethane derivs.
     396103-26-7P 396103-29-0P 396103-32-5P
     396103-36-9P 396103-38-1P 396103-40-5DP,
     reaction products with octanol 396103-40-5P 396103-42-7P
     396103-45-0P 396103-47-2P
     RL: IMF (Industrial manufacture); POF (Polymer in formulation); THU
     (Therapeutic use); BIOL (Biological study); PREP (Preparation); USES
     (Uses)
        (polyurethane dispersions in iso-Pr alc.-water system and
        manuf. for use in antimicrobial compns. and film-forming surgical
        drapes)
RN
     96-27-5 HCAPLUS
     1,2-Propanediol, 3-mercapto- (6CI, 8CI, 9CI) (CA INDEX NAME)
CN
        OH
HS-CH_2-CH-CH_2-OH
RN
     107-15-3 HCAPLUS
     1,2-Ethanediamine (9CI) (CA INDEX NAME)
CN
```

```
H2N-CH2-CH2-NH2
     79103-62-1 HCAPLUS
RN
     Desmodur W (9CI) (CA INDEX NAME)
CN
*** STRUCTURE DIAGRAM IS NOT AVAILABLE ***
    396103-26-7 HCAPLUS
RN
     1,3-Benzenedicarboxylic acid, 5-sulfo-, 1,3-dimethyl ester, sodium salt,
CN
    polymer with Desmodur I, 1,2-ethanediamine, .alpha.-hydro-.omega.-
    hydroxypoly(oxy-1,4-butanediyl), Kraton Liquid L 2203, 2-oxepanone,
     2,2'-oxybis[ethanol] and 2,4,7,9-tetramethyl-5-decyne-4,7-diol, block
     (9CI) (CA INDEX NAME)
     CM
          1
         202149-37-9
     CRN
     CMF Unspecified
     CCI PMS, MAN
*** STRUCTURE DIAGRAM IS NOT AVAILABLE ***
     CM
     CRN
         191617-94-4
         Unspecified
     CMF
     CCI PMS, MAN
*** STRUCTURE DIAGRAM IS NOT AVAILABLE ***
     CM
     CRN
          25190-06-1
     CMF
          (C4 H8 O)n H2 O
     CCI
          PMS
       - (CH<sub>2</sub>)<sub>4</sub>-0-
     CM
```

CRN 3965-55-7

CMF C10 H10 O7 S . Na

# Na

CM 5

CRN 502-44-3 CMF C6 H10 O2

CM 6

CRN 126-86-3 CMF C14 H26 O2

$$\begin{array}{c|c} OH & OH \\ \mid & \mid \\ i\text{-Bu-C-C} = \text{C-C-Bu-i} \\ \mid & \mid \\ Me & Me \end{array}$$

CM '

CRN 111-46-6 CMF C4 H10 O3

$${\tt HO-CH_2-CH_2-O-CH_2-CH_2-OH}$$

CM 8

CRN 107-15-3 CMF C2 H8 N2

```
H_2N - CH_2 - CH_2 - NH_2
     396103-29-0 HCAPLUS
RN
     Propanoic acid, 3-hydroxy-2-(hydroxymethyl)-2-methyl-, polymer with
CN
     Acclaim 3201, Desmodur W and 1,2-ethanediamine, block, potassium salt
     (9CI) (CA INDEX NAME)
          1
     CM
     CRN 396103-28-9
     CMF (C5 H10 O4 . C2 H8 N2 . Unspecified . Unspecified)x
     CCI PMS
     CDES 8:PM, BLOCK
          CM
               2
          CRN 188571-35-9
          CMF Unspecified
          CCI
               PMS, MAN
*** STRUCTURE DIAGRAM IS NOT AVAILABLE ***
          CM
               3
          CRN
              79103-62-1
          CMF
               Unspecified
          CCI MAN
*** STRUCTURE DIAGRAM IS NOT AVAILABLE ***
          CM
               4
          CRN 4767-03-7
          CMF C5 H10 O4
       Me
но-сн2-с-со2н
        {
m CH_2}-{
m OH}
          CM
               5
          CRN 107-15-3
          CMF C2 H8 N2
H_2N-CH_2-CH_2-NH_2
     396103-32-5 HCAPLUS
RN
     Propanoic acid, 3-hydroxy-2-(hydroxymethyl)-2-methyl-, polymer with
CN
     Desmodur W, 1,2-ethanediamine, .alpha.-hydro-.omega.-
     hydroxypoly[oxy(methyl-1,2-ethanediyl)] and .alpha.,.alpha.',.alpha.''-
     1,2,3-propanetriyltris[.omega.-hydroxypoly[oxy(methyl-1,2-ethanediyl)]],
     block, sodium salt (9CI) (CA INDEX NAME)
```

CM 1

CRN 396103-31-4

CMF (C5 H10 O4 . (C3 H6 O)n (C3 H6 O)n (C3 H6 O)n C3 H8 O3 . (C3 H6 O)n

H2 O . C2 H8 N2 . Unspecified)x

CCI PMS

CDES 8:PM, BLOCK

CM 2

CRN 79103-62-1

CMF Unspecified

CCI MAN

### \*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

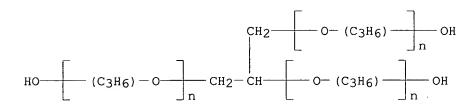
CM 3

CRN 25791-96-2

CMF (C3 H6 O)n (C3 H6 O)n (C3 H6 O)n C3 H8 O3

CCI IDS, PMS

CDES 8:ID



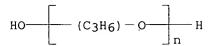
CM 4

CRN 25322-69-4

CMF (C3 H6 O)n H2 O

CCI IDS, PMS

CDES 8:ID



CM 5

CRN 4767-03-7

CMF C5 H10 O4

HO 
$$\left[ (C_3H_6) - O \right]_n$$
 H

```
CM
               5
         CRN 4767-03-7
         CMF C5 H10 O4
       Me
HO-CH2-C-CO2H
        сн2-он
         CM
               6
         CRN 107-15-3
         CMF C2 H8 N2
н2N-Сн2-Сн2-NН2
RN
     396103-38-1 HCAPLUS
    1,3-Benzenedicarboxylic acid, 5-sulfo-, 1,3-dimethyl ester, sodium salt,
CN
    polymer with Desmodur I, 1,2-ethanediamine, .alpha.-hydro-.omega.-
    hydroxypoly(oxy-1,2-ethanediyl), .alpha.-hydro-.omega.-
    hydroxypoly[oxy(methyl-1,2-ethanediyl)], Kraton Liquid L 2203 and Priplast
     3192, block (9CI) (CA INDEX NAME)
     CM
         1
    CRN 202149-37-9
     CMF
         Unspecified
    CCI
         PMS, MAN
*** STRUCTURE DIAGRAM IS NOT AVAILABLE ***
    CM
         2
         191617-94-4
    CRN
         Unspecified
    CMF
         PMS, MAN
    CCI
*** STRUCTURE DIAGRAM IS NOT AVAILABLE ***
          3
    CM
         157630-15-4
    CRN
         Unspecified
    CMF
         PMS, MAN
    CCI
*** STRUCTURE DIAGRAM IS NOT AVAILABLE ***
    CM
          4
    CRN 25322-69-4
     CMF
         (C3 H6 O)n H2 O
```

CCI IDS, PMS

CDES 8:ID

CM 5

CRN 25322-68-3 CMF (C2 H4 O)n H2 O · CCI PMS

$$HO = \begin{bmatrix} CH_2 - CH_2 - O \end{bmatrix}_n$$

CM 6

CRN 3965-55-7 CMF C10 H10 O7 S . Na

Na

CM 7

CRN 107-15-3 CMF C2 H8 N2

 $H_2N-CH_2-CH_2-NH_2$ 

RN 396103-40-5 HCAPLUS

CN 1,3-Benzenedicarboxylic acid, 5-sulfo-, 1,3-dimethyl ester, sodium salt, polymer with Desmodur I, 1,2-ethanediamine, .alpha.-hydro-.omega.-hydroxypoly(oxy-1,4-butanediyl), .alpha.-hydro-.omega.-hydroxypoly(oxy-1,2-ethanediyl), .alpha.-hydro-.omega.-hydroxypoly[oxy(methyl-1,2-ethanediyl)] and Kraton Liquid L 2203, block (9CI) (CA INDEX NAME)

CM 1

CRN 202149-37-9 CMF Unspecified CCI PMS, MAN

# \*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

CM 2

CRN 191617-94-4 CMF Unspecified CCI PMS, MAN

# \*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

CM 3

CRN 25322-69-4 CMF (C3 H6 O)n H2 O CCI IDS, PMS CDES 8:ID

CM 4

CRN 25322-68-3 CMF (C2 H4 O)n H2 O CCI PMS

CM 5

CRN 25190-06-1 CMF (C4 H8 O)n H2 O CCI PMS

$$HO = \begin{bmatrix} (CH_2)_4 - O \end{bmatrix}_n H$$

CM 6

CRN 3965-55-7 CMF C10 H10 O7 S . Na

Na

CM 7

CRN 107-15-3 CMF C2 H8 N2

 $H_2N-CH_2-CH_2-NH_2$ 

RN 396103-40-5 HCAPLUS

CN 1,3-Benzenedicarboxylic acid, 5-sulfo-, 1,3-dimethyl ester, sodium salt, polymer with Desmodur I, 1,2-ethanediamine, .alpha.-hydro-.omega.-hydroxypoly(oxy-1,4-butanediyl), .alpha.-hydro-.omega.-hydroxypoly(oxy-1,2-ethanediyl), .alpha.-hydro-.omega.-hydroxypoly[oxy(methyl-1,2-ethanediyl)] and Kraton Liquid L 2203, block (9CI) (CA INDEX NAME)

CM 1

CRN 202149-37-9 CMF Unspecified CCI PMS, MAN

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

CM 2

CRN 191617-94-4 CMF Unspecified CCI PMS, MAN

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

CM 3

CRN 25322-69-4 CMF (C3 H6 O)n H2 O CCI IDS, PMS CDES 8:ID

$$HO \longrightarrow (C3H6) - O \longrightarrow H$$

CM 4

CRN 25322-68-3

CMF (C2 H4 O)n H2 O

CCI PMS

$$HO \longrightarrow CH_2 - CH_2 - O \longrightarrow n$$

CM 5

CRN 25190-06-1

CMF (C4 H8 O)n H2 O

CCI PMS

CM 6

CRN 3965-55-7 CMF C10 H10 O7 S . Na

● Na

CM

CRN 107-15-3 CMF C2 H8 N2

H2N-CH2-CH2-NH2

RN 396103-42-7 HCAPLUS

CN 1,3-Benzenedicarboxylic acid, 5-sulfo-, 1,3-dimethyl ester, sodium salt,

polymer with Desmodur I, 1,2-ethanediamine, .alpha.-hydro-.omega.-hydroxypoly(oxy-1,2-ethanediyl), .alpha.-hydro-.omega.-hydroxypoly[oxy(methyl-1,2-ethanediyl)], Pripol 1009 and Pripol 2033, block (9CI) (CA INDEX NAME)

CM 1

CRN 202149-37-9 CMF Unspecified CCI PMS, MAN

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

CM 2

CRN 158516-85-9 CMF Unspecified

CCI MAN

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

CM 3

CRN 127290-22-6 CMF Unspecified

CCI MAN

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

CM 4

CRN 25322-69-4 CMF (C3 H6 O)n H2 O CCI IDS, PMS CDES 8:ID

CM 5

CRN 25322-68-3 CMF (C2 H4 O)n H2 O CCI PMS

CM 6

CRN 3965-55-7 CMF C10 H10 O7 S . Na

Na

CM 7

CRN 107-15-3 CMF C2 H8 N2

 $_{\rm H_2N^-\,CH_2^-\,CH_2^-\,NH_2}$ 

RN 396103-45-0 HCAPLUS

CN 1,3-Benzenedicarboxylic acid, 5-sulfo-, 1,3-dimethyl ester, sodium salt, polymer with N,N-bis(2-aminoethyl)-1,2-ethanediamine, Desmodur I, .alpha.-hydro-.omega.-hydroxypoly(oxy-1,2-ethanediyl), .alpha.-hydro-.omega.-hydroxypoly[oxy(methyl-1,2-ethanediyl)], Kraton Liquid L 2203 and Priplast 3192, block (9CI) (CA INDEX NAME)

CM 1

CRN 202149-37-9 CMF Unspecified CCI PMS, MAN

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

CM 2

CRN 191617-94-4 CMF Unspecified CCI PMS, MAN

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

CM 3

CRN 157630-15-4 CMF Unspecified CCI PMS, MAN

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

CM 4

CRN 25322-69-4

CMF (C3 H6 O)n H2 O CCI IDS, PMS CDES 8:ID

$$HO = \begin{bmatrix} (C3H6) - O \end{bmatrix}_n H$$

CM 5

CRN 25322-68-3 CMF (C2 H4 O)n H2 O CCI PMS

$$HO \longrightarrow CH_2 - CH_2 - O \longrightarrow n$$

CM 6

CRN 4097-89-6 CMF C6 H18 N4

$$\begin{array}{c} \text{CH}_2-\text{CH}_2-\text{NH}_2 \\ | \\ \text{H}_2\text{N}-\text{CH}_2-\text{CH}_2-\text{N}-\text{CH}_2-\text{CH}_2-\text{NH}_2 \end{array}$$

CM 7

CRN 3965-55-7 CMF C10 H10 O7 S . Na

# Na

RN 396103-47-2 HCAPLUS

CN 1,3-Benzenedicarboxylic acid, 5-sulfo-, 1,3-dimethyl ester, sodium salt, polymer with Desmodur I, 1,2-ethanediamine, .alpha.-hydro-.omega.-hydroxypoly(oxy-1,4-butanediyl), .alpha.-hydro-.omega.-hydroxypoly(oxy-1,2-

ethanediyl), .alpha.-hydro-.omega.-hydroxypoly[oxy(methyl-1,2-ethanediyl)], Kraton Liquid L 2203 and 2,4,7,9-tetramethyl-5-decyne-4,7-diol, block (9CI) (CA INDEX NAME)

CM 1

CRN 202149-37-9 CMF Unspecified CCI PMS, MAN

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

CM 2

CRN 191617-94-4 CMF Unspecified CCI PMS, MAN

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

CM 3

CRN 25322-69-4 CMF (C3 H6 O)n H2 O CCI IDS, PMS CDES 8:ID

$$HO \longrightarrow (C_3H_6) - O \longrightarrow n$$

CM 4

CRN 25322-68-3 CMF (C2 H4 O)n H2 O CCI PMS

$$HO \longrightarrow CH_2 - CH_2 - O \longrightarrow H$$

CM 5

CRN 25190-06-1 CMF (C4 H8 O)n H2 O CCI PMS

HO (CH<sub>2</sub>)<sub>4</sub> - O 
$$\frac{1}{n}$$
 H

CM 6

CRN 3965-55-7

CMF C10 H10 O7 S . Na

Na

CM 7

CRN 126-86-3 CMF C14 H26 O2

CM 8

CRN 107-15-3 CMF C2 H8 N2

H2N-CH2-CH2-NH2

IT 538-23-8, Crodamol GTCC

RL: MOA (Modifier or additive use); USES (Uses)
 (polyurethane dispersions in iso-Pr alc.-water system and
 manuf. for use in antimicrobial compns. and film-forming surgical
 drapes)

RN 538-23-8 HCAPLUS

CN Octanoic acid, 1,2,3-propanetriyl ester (9CI) (CA INDEX NAME)

RN 64-17-5 HCAPLUS

CN Ethanol (9CI) (CA INDEX NAME)

$${\tt H3C-CH2-OH}$$

RN 67-63-0 HCAPLUS CN 2-Propanol (9CI) (CA INDEX NAME)

RN 71-23-8 HCAPLUS CN 1-Propanol (9CI) (CA INDEX NAME)

 $H_3C-CH_2-CH_2-OH$ 

REFERENCE COUNT: 5 THERE ARE 5 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

# => d ibib abs hitstr 1

L45 ANSWER 1 OF 3 HCAPLUS COPYRIGHT 2002 ACS 1998:479570 HCAPLUS ACCESSION NUMBER:

DOCUMENT NUMBER:

129:126909

TITLE:

Cationic film-forming polymer compositions

for topical delivery of agents to the skin and hair

INVENTOR(S):

Samour, Carlos; Krauser, Scott F.

PATENT ASSIGNEE(S):

Macrochem Corp., USA

SOURCE:

PCT Int. Appl., 49 pp.

CODEN: PIXXD2

DOCUMENT TYPE:

Patent

LANGUAGE:

English

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

PA	PATENT NO.				KIND DATE				APPLICATION NO.						DATE				
WO	9828355			A:	 1	19980702			WO 1997-US23999					1997	1223				
	W:	ΑU,	BR,	CA,	CN,	CZ,	HU,	IL,	JP,	KR,	MX,	PL,	RU,	US,	AM,	ΑZ,	BY,		
		KG,	ΚZ,	MD,	RU,	ТJ,	TM												
	RW:	ΑT,	BE,	CH,	DE,	DK,	ES,	FΙ,	FR,	GB,	GR,	ΙE,	ΙT,	LU,	MC,	NL,	PT,	SE	
								US 1996-771990											
US	5906822			Α		19990525			Ū	S 19	97-9	3764	9	1997	0925				
AU	9859036			A1 19980717			Α	U 19	98-5	9036		1997	1223						
EP	946621			A1 1999100		1006		E	P 19	97-9	5486	8	1997	1223					
	R:	ΑT,	BE,	CH,	DE,	DK,	ES,	FR,	GB,	GR,	ΙT,	LI,	LU,	NL,	SE,	MC,	PT,		
		ΙE,																	
JP 2001507386 T2						2001	0605		J	P 19	98-5	2908	8	1997	1223				
PRIORITY APPLN. INFO.:								US 1	996-	7719	90	Α	1996	1223					
									US 1	995-	4964	13	В3	1995	0629				
					1	WO 1	997-	US23	999	M	1997	1223							

AB A mixt. of lipophilic and amphiphilic/hydrophilic film-forming polymers R(CO)mY1(CH2CHR10)n[CONHZNHCO(OCH2CHR2)n'Y2(CH2CHR20)n'']pCONHZNH CO(OCH2CHR1)nY1(CO)mR where R represents (i) an alkyl, alkenyl or alkaryl group of from 1 to 30 carbon atoms or (ii) a polypropylene oxide group or polybutylene oxide group; R1 and R2, each, independently, represent a hydrogen atom, or a Me or Et group; Z represents a divalent linking hydrocarbyl group; m is 0 or 1; when m = 0, Y1 represents 0, NR3 or N+R3R4X- and when m=1 Y1 represents O; Y2 represents O, NR3, N+R3R4X-, (R3NCH2CH2)t, or [(R3R4N+CH2CH2)X-]t; R3 and R4, independently, represent C1-C22 alkyl; X represents an anion; t is a pos. integer; n and n', are each, independently, a pos. no. and p is >0, provide a skin substantive, occlusive moisture barrier covering for skin or hair. When Y1 and Y2 are O the polymers are nonionic in character when Y1 or Y2 represents NR3 or N+R3R4X- the polymers are cationic in character. The compns. have a moisturizing effect by reducing transepidermal water loss. The lipophilic or amphiphilic or hydrophilic film-forming polymers may be used individually or as mixts. as a delivery system for delivering pharmacol. or cosmetic agents to the skin or hair. A polymer was prepd. fro dicyclomethane 4,4'-diisocyanate, Ethoquad 18/25, and polyoxyethylene stearyl ether. A moisturizing lotion was prepd. contg. this polymer.

ΤТ 210243-08-6P

> RL: BUU (Biological use, unclassified); SPN (Synthetic preparation); THU (Therapeutic use); BIOL (Biological study); PREP (Preparation); USES (Uses)

(cationic film-forming polymer compns. for topical delivery to the skin and hair)

210243-08-6 HCAPLUS RN

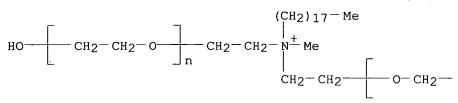
Poly(oxy-1,2-ethanediyl), .alpha.,.alpha.'-[(methyloctadecyliminio)di-2,1-CN

ethanediyl]bis[.omega.-hydroxy-, chloride, polymer with 1,1'-methylenebis[4-isocyanatocyclohexane] and .alpha.-octadecyl-.omega.-hydroxypoly(oxy-1,2-ethanediyl) (9CI) (CA INDEX NAME)

CM :

CRN 28724-32-5 CMF (C2 H4 O)n (C2 H4 O)n C23 H50 N O2 . C1 CCI PMS

PAGE 1-A



● Cl-

PAGE 1-B

CM 2

CRN 9005-00-9 CMF (C2 H4 O)n C18 H38 O CCI PMS

CM 3

CRN 5124-30-1 CMF C15 H22 N2 O2

### => d ind

L45 ANSWER 1 OF 3 HCAPLUS COPYRIGHT 2002 ACS

IC ICM C08G018-48

ICS C08G018-50; A61K031-27; A61L015-42

CC 62-4 (Essential Oils and Cosmetics)

Section cross-reference(s): 63

ST cationic **film** forming polymer skin hair; polyoxyalkylene

urethane topical delivery skin hair

IT Acne

Antibiotics

Antitumor agents

Cosmetics

### Fungicides

Hair preparations

Local anesthetics

Psoriasis

Sunscreens

Topical drug delivery systems

(cationic film-forming polymer compns. for topical delivery

to the skin and hair)

IT Polyoxyalkylene-polyurethanes

RL: BUU (Biological use, unclassified); THU (Therapeutic use); BIOL

(Biological study); USES (Uses)

(cationic film-forming polymer compns. for topical delivery

to the skin and hair)

IT Hormones (animal), biological studies

RL: THU (Therapeutic use); BIOL (Biological study); USES (Uses)

(cationic film-forming polymer compns. for topical delivery

to the skin and hair)

## IT 210243-08-6P

RL: BUU (Biological use, unclassified); SPN (Synthetic preparation); THU (Therapeutic use); BIOL (Biological study); PREP (Preparation); USES

(Uses)

(cationic **film**-forming polymer compns. for topical delivery

to the skin and hair)

#### => d ibib abs hitstr 2

L45 ANSWER 2 OF 3 HCAPLUS COPYRIGHT 2002 ACS 1992:470926 HCAPLUS ACCESSION NUMBER:

DOCUMENT NUMBER:

117:70926

TITLE:

Manufacture of particulate polymers

INVENTOR(S):

Kataoka, Hironori; Jinno, Kazuto; Yamashita, Tokiko

APPLICATION NO. DATE

PATENT ASSIGNEE(S): Daiichi Kogyo Seiyaku K. K., Japan

Jpn. Kokai Tokkyo Koho, 8 pp.

SOURCE:

DOCUMENT TYPE:

CODEN: JKXXAF

Patent Japanese

LANGUAGE:

KIND DATE

FAMILY ACC. NUM. COUNT:

PATENT NO.

PATENT INFORMATION:

	JP 04076016	A2 19920310	JP 1990-189965	19900718
AB	The title partice	lates, useful as	matting agents for c	oatings,
	additives for add	nesives and cosmet	tics, and for manuf.	of
	release-controlle	ed drugs or pesti	cides, are prepd. by	emulsifying a mixt.
	of diisocyanates,	polyisocyanates	, urethane prepolymer	s, surfactants
	bearing .gtoreq.2	2 OH groups, and a	a protective colloid,	then curing the
	mixt. Thus, stin	rring 100 g 3:1 t:	rimethylolpropane-TDI	prepolymer with 30
	g C2H4O-C3H6O blo	ock copolymer, add	ding 700 g 2% aq. pol	y(acrylic acid) Na
	salt soln., and s	stirring at 40.de	gree. for 5 h gave pa	rticles with av.

diam. 110 .mu.m. IT 142807-54-3P

RL: PREP (Preparation)

(particles, with uniform size, manuf. of)

RN 142807-54-3 HCAPLUS

1-Dodecanaminium, N-(4-carboxybutyl)-2-hydroxy-N-(2-hydroxydodecyl)-N-(2-CN hydroxyethyl)-, inner salt, polymer with 1,3-diisocyanatomethylbenzene and 2-ethyl-2-(hydroxymethyl)-1,3-propanediol (9CI) (CA INDEX NAME)

CM

CRN 134845-59-3 CMF C31 H63 N O5

2 CM

CRN 26471-62-5 CMF C9 H6 N2 O2

CCI IDS CDES 8:ID

(particles, prepn. of, for suede-finish coatings)

ΙT

142816-66-8P

RL: PREP (Preparation)

IT 101-68-8DP, polymers with castor oil monosulfate 4098-71-9DP, polymers with polypropylene glycol glycerol ether and castor oil 25791-96-2DP, polymers with isophorone diisocyanate and castor oil 25854-16-4DP, Xylylene diisocyanate, polymers with castor oil monosulfate 142708-71-2P 142708-72-3P 142807-54-3P RL: PREP (Preparation)

(particles, with uniform size, manuf. of)

IT 50-78-2, Aspirin 541-91-3, Muscone 9001-22-3, .beta.-Glucosidase RL: USES (Uses) (polyurethane particles contg., manuf. of)

=> d ibib abs hitstr 1

L13 ANSWER 1 OF 7 HCAPLUS COPYRIGHT 2002 ACS ACCESSION NUMBER:

1999:129295 HCAPLUS

DOCUMENT NUMBER:

130:325648

TITLE:

Surface modification of segmented polyurethanes by grafting methacrylates and phosphatidylcholine polar

headgroups to improve hemocompatibility

AUTHOR(S): CORPORATE SOURCE:

Li, Yu-Jun; Hanada, Tomohiro; Nakaya, Tadao Department of Bio-applied Chemistry Faculty of Engineering, Osaka City University, Sumiyoshi-ku

Osaka, 5588585, Japan

SOURCE:

PUBLISHER:

Chem. Mater. (1999), 11(3), 763-770

CODEN: CMATEX; ISSN: 0897-4756

American Chemical Society

DOCUMENT TYPE:

Journal

English LANGUAGE:

A new type of phosphatidylcholine-contg. segmented polyurethane surface was produced by grafting various methacrylates and phosphatidylcholine polar headgroups to a vinyl-group-contg. segmented polyurethane (V-SPU) using AIBN as a radical initiator. 1,4-Butanediol as chain extender was used to synthesize the V-SPU, which is based on MDI and vinyl-group-contg. polybutadiene diol. Several methacrylates, such as Me methacrylate (MMA), Bu methacrylate, stearyl methacrylate, and a phosphatidylcholine polar-headgroup-contg. vinyl monomer, 2-(methacryloyloxy)ethyl 2-(trimethylammonio)ethyl phosphate (MTP), were grafted to the V-SPU. The bulk characteristics of the grafted V-SPUs were investigated by IR spectroscopy, viscosity, and gel-permeation chromatog. measurements. Mech. properties of MMA-grafted V-SPU were measured by dynamic viscoelasticity and tensile measurements. The phosphatidylcholine polar headgroups were oriented on the surface of these materials, as revealed by attenuated total reflectance FTIR, ESCA, and contact angle measurements. The MTP-grafted SPU surfaces showed slightly decreased water contact angles, also indicating that hydrophilic phosphatidylcholine polar headgroups are present at the surface. The hemocompatibility in vitro was evaluated with platelet-rich plasma contact tests and viewed by SEM using ungrafted V-SPU as a ref. It was found that fewer platelets adhered to the modified surfaces and showed less shape variation than to the unmodified V-SPU. Platelet adhesion to MTP-grafted polymers was inhibited 88-95% compared with unmodified V-SPU.

223801-76-1P 223801-78-3P, Butadiene-1,4-butanediol-ITbutyl methacrylate-MDI-2-(methacryloyloxy)ethyl 2-(trimethylammonio)ethyl phosphate block graft copolymer 223801-80-7P, Butadiene-1, 4-butanediol-MDI-2-(methacryloyloxy) ethyl 2-(trimethylammonio)ethyl phosphate-stearyl methacrylate block graft

RL: BAC (Biological activity or effector, except adverse); SPN (Synthetic preparation); BIOL (Biological study); PREP (Preparation) (prepn. and hemocompatibility of)

223801-76-1 HCAPLUS RN

CN 3,5,8-Trioxa-4-phosphaundec-10-en-1-aminium, 4-hydroxy-N,N,N,10tetramethyl-9-oxo-, inner salt, 4-oxide, polymer with 1,3-butadiene, 1,4-butanediol, 1,1'-methylenebis[4-isocyanatobenzene] and methyl 2-methyl-2-propenoate, block, graft (9CI) (CA INDEX NAME)

CM 1

copolymer

CRN 67881-98-5 CMF C11 H22 N O6 P

CM 2

CRN 110-63-4 CMF C4 H10 O2

 $HO-(CH_2)_4-OH$ 

CM 3

CRN 106-99-0 CMF C4 H6

$$H_2C = CH - CH = CH_2$$

CM 4

CRN 101-68-8 CMF C15 H10 N2 O2

CM 5

CRN 80-62-6 CMF C5 H8 O2

RN 223801-78-3 HCAPLUS

CN 3,5,8-Trioxa-4-phosphaundec-10-en-1-aminium, 4-hydroxy-N,N,N,10-tetramethyl-9-oxo-, inner salt, 4-oxide, polymer with 1,3-butadiene, 1,4-butanediol, butyl 2-methyl-2-propenoate and 1,1'-methylenebis[4-isocyanatobenzene], block, graft (9CI) (CA INDEX NAME)

CM 1

CRN 67881-98-5 CMF C11 H22 N O6 P

CM 2

CRN 110-63-4 CMF C4 H10 O2

 $HO-(CH_2)_4-OH$ 

CM 3

CRN 106-99-0 CMF C4 H6

$$H_2C = CH - CH = CH_2$$

CM 4

CRN 101-68-8 CMF C15 H10 N2 O2

CM 5

CRN 97-88-1 CMF C8 H14 O2

RN 223801-80-7 HCAPLUS

CN 3,5,8-Trioxa-4-phosphaundec-10-en-1-aminium, 4-hydroxy-N,N,N,10-tetramethyl-9-oxo-, inner salt, 4-oxide, polymer with 1,3-butadiene,

- 1,4-butanediol, 1,1'-methylenebis[4-isocyanatobenzene] and octadecyl 2-methyl-2-propenoate, block, graft (9CI) (CA INDEX NAME)
- CM 1
- CRN 67881-98-5 CMF C11 H22 N O6 P
- - CM 2
  - CRN 32360-05-7 CMF C22 H42 O2
- $$^{\rm O}_{\rm H_2}$$  Me $^-$  (CH2)  $_{17}$  O– C– C– Me
  - CM 3
  - CRN 110-63-4 CMF C4 H10 O2
- $^{\rm HO-}$  (CH<sub>2</sub>)<sub>4</sub> $^{\rm -}$ OH
  - CM 4
  - CRN 106-99-0 CMF C4 H6
- $H_2C \longrightarrow CH CH \longrightarrow CH_2$ 
  - CM 5
  - CRN 101-68-8
  - CMF C15 H10 N2 O2
- OCN CH2 NCO

REFERENCE COUNT:

THERE ARE 27 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

#### => d ibib abs hitstr 2

L13 ANSWER 2 OF 7 HCAPLUS COPYRIGHT 2002 ACS ACCESSION NUMBER:

1998:392152 HCAPLUS

DOCUMENT NUMBER:

129:58850

TITLE:

Extended wear ophthalmic lens containing oxyperm

macromer/ionoperm monomer copolymer

INVENTOR(S):

Nicolson, Paul Clement; Baron, Richard Carlton; Chabrecek, Peter; Court, John; Domschke, Angelika; Griesser, Hans Jorg; Ho, Arthur; Hopken, Jens; Laycock, Bronwyn Glenice; Liu, Qin; Lohmann, Dieter; Meijs, Gordon Francis; Papaspiliotopoulos, Eric; Riffle, Judy Smith; Schindhelm, Klaus; Sweeney, Deborah; Terry, Wilson Leonard, Jr.; Vogt, Jurgen; Winterton, Lynn Cook

PATENT ASSIGNEE(S):

Ciba Vision Corp., USA; Commonwealth Scientific and

Industrial Research Organisation

SOURCE:

U.S., 38 pp. Cont.-in-part of U.S. 301,166, abandoned. CODEN: USXXAM

DOCUMENT TYPE:

Patent

LANGUAGE:

English

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

PATENT NO.								APPLICATION NO.					DATE				
US 5760100 A				19980602				US 1995-569816					19951208				
			В	1					· · · · · · · · · · · · · · · · · · ·								
US 5760100 TW 393498			B 20000611					TW 1995-84108480									
CA 2213357			AA 19961010										19960322				
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ΑU	AU 704749 B			2	19990506												
ΕP	EP 819258		Α	A1 19980121				EP 1996-908116 19960322									
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AU 9711399
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EP 1996-939249
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EP 865615
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        IE, FI
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. 19971118 NO 9704585 Α NO 1997-4584 19971003 US 6043328 20000328 Α US 1997-952416 19971117 AU 9935828 . A1 19990916 AU 1999-35828 19990622 PRIORITY APPLN. INFO.: US 1994-301166 B2 19940906 EP 1995-810221 А 19950404 CH 1995-1496 Α 19950519 CH 1995-1476 А 19950518 US 1995-569816 Α 19951208 AU 1996-51478 A3 19960322 EP 1996-908116 19960322 ДЗ WO 1996-EP1255 W 19960322 WO 1996-EP1265 W 19960322 19960507 WO 1996-EP1888 W EP 1996-810643 Α 19960930 US 1996-27736P Ρ 19961003 WO 1996-EP5326 W 19961202 WO 1996-IB1368 W 19961206

AB An ophthalmic lens having high oxygen permeability and ion or water permeability suited for extended-wear at least one day comprises a copolymer of a oxyperm macromer and an ionoperm monomer. Thus, 100 g poly(dimethylsiloxane) dialkanol was reacted in sequence with 21.2 g IPDI, 610 g poly(ethylene glycol), 10.4 g isocyanate methacrylate to give a macromer, 180 g of which was polymd. with 3-methacryloxypropyltris (trimethylsiloxy)silane 15 g, ethylene glycol dimethacrylate 1 g, and 2-hydroxyethyl methacrylate 4 g in a polypropylene mold to form a contact lens, showing Hydrodell water permeability coeff. 0.71x10-6 cm2/s and moving of the lens on the eye.

## IT 208589-57-5P

RL: IMF (Industrial manufacture); THU (Therapeutic use); BIOL (Biological study); PREP (Preparation); USES (Uses)

(extended wear ophthalmic lens contg. oxyperm macromer/ionoperm monomer copolymer)

RN 208589-57-5 HCAPLUS

CN 1-Propanaminium, 2-hydroxy-N,N,N-trimethyl-3-[(2-methyl-1-oxo-2-propenyl)oxy]-, chloride, polymer with N,N-dimethyl-2-propenamide, Fomblin Z-DOL, .alpha.-[[3-(2-hydroxyethoxy)propyl]dimethylsilyl]-.omega.-[[[3-(2-hydroxyethoxy)propyl]dimethylsilyl]oxy]goly[oxy(dimethylsilylene)], 2-isocyanatoethyl 2-methyl-2-propenoate, 5-isocyanato-1-(isocyanatomethyl)-1,3,3-trimethylcyclohexane and 3-[3,3,3-trimethyl-1,1-bis[(trimethylsilyl)oxy]disiloxanyl]propyl 2-methyl-2-propenoate (9CI) (CA INDEX NAME)

CM 1

CRN 156327-07-0

CMF (C2 H6 O Si)n C14 H34 O5 Si2

CCI PMS

PAGE 1-B

— cн<sub>2</sub>— он

CM 2

CRN 107852-51-7 CMF Unspecified CCI PMS, MAN

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

CM 3

CRN 30674-80-7 CMF C7 H9 N O3

CM 4

CRN 17096-07-0 CMF C16 H38 O5 Si4

CM 5

CRN 13052-11-4 CMF C10 H20 N O3 . C1

Cl-

CM 6

CRN 4098-71-9 CMF C12 H18 N2 O2

CM 7

CRN 2680-03-7 CMF C5 H9 N O

### => d ibib abs hitstr 3

L13 ANSWER 3 OF 7 HCAPLUS COPYRIGHT 2002 ACS ACCESSION NUMBER: 1998:361027 HCAPLUS

DOCUMENT NUMBER: 129:55279

Hydrophilic, antisoiling, and biocompatible resins and TITLE:

medical materials containing them

Koinuma, Yasuyoshi; Inomata, Kiyoshi; Nakabayashi, INVENTOR(S):

Nobuo; Ishihara, Kazuhiko

PATENT ASSIGNEE(S): Nippon Oil and Fats Co., Ltd., Japan; Nakabayashi,

Norio; Ishihara, Kazuhiko; Foundation for Scientific

Technology Promotion

SOURCE: Jpn. Kokai Tokkyo Koho, 7 pp.

CODEN: JKXXAF

DOCUMENT TYPE:

Patent

LANGUAGE:

Japanese

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

KIND DATE APPLICATION NO. DATE PATENT NO. JP 10152533 A2 19980609 JP 1996-312102 19961122

AΒ The resins having excellent mech. strength are obtained by polymg. 1-99% polymers having (meth)acrylate ester units CH2CR1CO2(CH2CHR2O)nP(O)(O-)OCH2CH2N+Me3 (R1, R2 = H, Me; n = 1-10) and 1-99% room-temp.-liq. polymerizable compds. Thus, 2-methacryloyloxyethyl 2-(trimethylammonio)ethyl phosphate was polymd. in EtOH in the presence of AIBN to give 92% polymer, 5 g of which was further polymd. with 10 g Me methacrylate at 60.degree. for 20 h in EtOH in the presence of AIBN, pptd. in (Me2CH)2O, filtered, dried, dissolved in EtOH, and cast on a hot plate to give a sheet showing contact angle 36 degree initially and 35 after rubbing by wrapping paper with particle size 3 .mu.m for 10 times, albumin adsorption 2.7 .mu.g/cm2 initially and 2.9 after the rubbing, and good antithrombogenic property.

#### ΙT 208663-43-8P

RL: IMF (Industrial manufacture); PRP (Properties); TEM (Technical or engineered material use); THU (Therapeutic use); BIOL (Biological study);
PREP (Preparation); USES (Uses)

(phosphorylcholine-contg. (meth)acrylate polymers with good antisoiling property and biocompatibility) 208663-43-8 HCAPLUS

RN

3,5,8-Trioxa-4-phosphaundec-10-en-1-aminium, 4-hydroxy-N,N,N,10-CN tetramethyl-9-oxo-, inner salt, 4-oxide, polymer with 1,3bis(isocyanatomethyl)benzene, 1,4-butanediol and 2-hydroxyethyl 2-methyl-2-propenoate (9CI) (CA INDEX NAME)

CM1

CRN 67881-98-5 CMF C11 H22 N O6 P

CM 2

CRN 3634-83-1 CMF C10 H8 N2 O2

CM 3

CRN 868-77-9 CMF C6 H10 O3

CM 4

CRN 110-63-4 CMF C4 H10 O2

 $^{\rm HO-}$  (CH<sub>2</sub>)<sub>4</sub> $^{\rm -}$ OH

### => d ibib abs hitstr 4

L13 ANSWER 4 OF 7 HCAPLUS COPYRIGHT 2002 ACS ACCESSION NUMBER: 1997:602699 HCAPLUS

DOCUMENT NUMBER: 127:263215

TITLE: Polyurethanes containing phospholipid analogs and

their manufacture

INVENTOR(S): Sugiyama, Kazuo; Fukuchi, Mikio; Akashi, Mitsuru;

Kishida, Akio; Kadoma, Yoshihito

PATENT ASSIGNEE(S): Nippon Oil and Fats Co., Ltd., Japan

SOURCE: Jpn. Kokai Tokkyo Koho, 10 pp.

CODEN: JKXXAF

DOCUMENT TYPE: Patent LANGUAGE: Japanese

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE

JP 09235342 A2 19970909 JP 1996-43241 19960229

The polyurethanes, useful for medical goods, comprise repeating units of (CONHZNHCO2QCH2O)n(CONHZNHCO2R)m [Q = CHCH2S[CH2C(Me)(CO2Me)]p[CH2C(Me)[CO2CH2CH2OP(O)(O-)OCH2CH2N+Me3]]q; Z = phenylene substituted by 0-4 C1-4 alkyl groups, linear or branched (un)satd. C1-12 alkylene-contg. alkylenediphenyl; R = (AO)u; A = C2-4 alkylene; u = 10-100; m = 0-0.99; n = 0.01-1; m + n = 1; p = 0-0.99; q = 0.01-1; p + q = 1]. Thus, polymn. of 2-methacryloyloxyethyl phosphorylcholine 5, MMA 6.78, and 1-thioglycerol 0.09 g in the presence of AIBN gave a macromer (no.-av. mol. wt. 22,000), which was reacted with 4,4'-diphenylmethane diisocyanate and polyethylene glycol to give a polymer.

IT 196314-43-9P 196314-46-2P 196314-50-8P 196314-52-0P

RL: IMF (Industrial manufacture); PRP (Properties); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (polyurethanes contg. phospholipid analogs and their manuf.)

RN 196314-43-9 HCAPLUS

CN 3,5,8-Trioxa-4-phosphaundec-10-en-1-aminium, 4-hydroxy-N,N,N,10-tetramethyl-9-oxo-, inner salt, 4-oxide, telomer with 3-mercapto-1,2-propanediol and methyl 2-methyl-2-propenoate, polymer with .alpha.-hydro-.omega.-hydroxypoly(oxy-1,2-ethanediyl) and 1,1'-methylenebis[4-isocyanatobenzene], block, graft (9CI) (CA INDEX NAME)

CM 1

CRN 25322-68-3 CMF (C2 H4 O)n H2 O CCI PMS

$$HO = \begin{bmatrix} CH_2 - CH_2 - O \end{bmatrix}_n H$$

CM 2

CRN 101-68-8

CMF C15 H10 N2 O2

CM 3

CRN 196314-38-2

CMF (C11 H22 N O6 P . C5 H8 O2)x . C3 H8 O2 S

CM 4

CRN 96-27-5 CMF C3 H8 O2 S

CM 5

CRN 67882-00-2

CMF (C11 H22 N O6 P . C5 H8 O2) $\times$ 

CCI PMS

CM 6

CRN 67881-98-5 CMF C11 H22 N O6 P

CM 7

CRN 80-62-6 CMF C5 H8 O2

RN 196314-46-2 HCAPLUS

CN 3,5,8-Trioxa-4-phosphaundec-10-en-1-aminium, 4-hydroxy-N,N,N,10-

tetramethyl-9-oxo-, inner salt, 4-oxide, telomer with 3-mercapto-1,2-propanediol and methyl 2-methyl-2-propenoate, polymer with .alpha.-hydro-.omega.-hydroxypoly[oxy(methyl-1,2-ethanediyl)] and 1,1'-methylenebis[4-isocyanatobenzene], block, graft (9CI) (CA INDEX NAME)

CM 1

CRN 25322-69-4 CMF (C3 H6 O)n H2 O CCI IDS, PMS CDES 8:ID

CM 2

CRN 101-68-8 CMF C15 H10 N2 O2

CM 3

CRN 196314-38-2 CMF (C11 H22 N O6 P . C5 H8 O2)× . C3 H8 O2 S

CM 4

CRN 96-27-5 CMF C3 H8 O2 S

$$\begin{array}{c} \text{OH} \\ \mid \\ \text{HS-CH}_2\text{-CH-CH}_2\text{-OH} \end{array}$$

CM 5

CRN 67882-00-2

CMF (C11 H22 N O6 P . C5 H8 O2)x

CCI PMS

CM 6

CRN 67881-98-5 CMF C11 H22 N O6 P

CM 7

CRN 80-62-6 CMF C5 H8 O2

RN 196314-50-8 HCAPLUS

CN 3,5,8-Trioxa-4-phosphaundec-10-en-1-aminium, 4-hydroxy-N,N,N,10-tetramethyl-9-oxo-, inner salt, 4-oxide, telomer with 3-mercapto-1,2-propanediol and methyl 2-methyl-2-propenoate, polymer with 1,3-diisocyanatomethylbenzene and .alpha.-hydro-.omega.-hydroxypoly(oxy-1,2-ethanediyl), block, graft (9CI) (CA INDEX NAME)

CM 1

CRN 26471-62-5 CMF C9 H6 N2 O2 CCI IDS CDES 8:ID

D1-Me

CM 2

CRN 25322-68-3 CMF (C2 H4 O)n H2 O CCI PMS

CM 3

CRN 196314-38-2

CMF (C11 H22 N O6 P . C5 H8 O2)x . C3 H8 O2 S

CM 4

CRN 96-27-5 CMF C3 H8 O2 S

 $\begin{array}{c} \text{OH} \\ | \\ \text{HS-CH}_2\text{--CH-CH}_2\text{--OH} \end{array}$ 

CM 5

CRN 67882-00-2

CMF (C11 H22 N O6 P . C5 H8 O2) x

CCI PMS

CM 6

CRN 67881-98-5 CMF C11 H22 N O6 P

CM 7

CRN 80-62-6 CMF C5 H8 O2

H<sub>2</sub>C O || || Me-C-C-OMe

RN 196314-52-0 HCAPLUS

CN 3,5,8-Trioxa-4-phosphaundec-10-en-1-aminium, 4-hydroxy-N,N,N,10-tetramethyl-9-oxo-, inner salt, 4-oxide, telomer with 3-mercapto-1,2-propanediol and methyl 2-methyl-2-propenoate, polymer with 1,3-diisocyanatomethylbenzene and .alpha.-hydro-.omega.-hydroxypoly[oxy(methyl-1,2-ethanediyl)], block, graft (9CI) (CA INDEX NAME)

CM 1

CRN 26471-62-5 CMF C9 H6 N2 O2 CCI IDS CDES 8:ID

D1-Me

2 CM

CRN 25322-69-4

CMF (C3 H6 O)n H2 O CCI IDS, PMS CDES 8:ID

$$HO = \begin{bmatrix} (C3H_6) - O \end{bmatrix}_n H$$

CM 3

CRN 196314-38-2

(C11 H22 N O6 P . C5 H8 O2)x . C3 H8 O2 S CMF

CM

96-27-5 CRN C3 H8 O2 S CMF

$$\begin{array}{c} \text{OH} \\ | \\ \text{HS--CH}_2\text{--CH--CH}_2\text{--OH} \end{array}$$

CM

67882-00-2 CRN

(C11 H22 N O6 P . C5 H8 O2)  $\times$ CMF

CCI PMS

CM

CRN 67881-98-5

CMF C11 H22 N O6 P

$$\begin{array}{c|c} ^{H_2C} & \text{O} \\ \parallel & \parallel \\ \text{Me-} & \text{C-} & \text{C-} & \text{OMe} \end{array}$$

=> d ibib abs hitstr 5

L13 ANSWER 5 OF 7 HCAPLUS COPYRIGHT 2002 ACS ACCESSION NUMBER: 1994:90732 HCAPLUS

DOCUMENT NUMBER: 120:90732

TITLE: Microcapsules having excellent mechanical strength and

manufacture thereof

INVENTOR(S): Inaba, Yoshihiro; Mikami, Takeshi

PATENT ASSIGNEE(S): Fuji Xerox Co Ltd, Japan

SOURCE: Jpn. Kokai Tokkyo Koho, 12 pp.

CODEN: JKXXAF

DOCUMENT TYPE: Patent LANGUAGE: Japanese

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 05061237	A2	19930312	JP 1992-20718	19920110
US 5336581	Α	19940809	US 1992-887049	19920526
PRIORITY APPLN. INFO.:	:	J	P 1991-153829	19910530
		J	P 1992-20718	19920110

AB Microcapsules are manufd. via the steps of effecting reaction of a 1st monomer with a 2nd monomer (at least either of which contains azo group capable of initiating polymn.) using internal in-situ polymn. to form capsule shells contg. the azo and a polymer structure selected from polyureas, polyurethanes, polyamides, polyesters, and epoxy resins, and effecting vinyl polymn. initiated by decompn. of the azo group to attach a vinyl monomer to the shells (i.e., forming a block and/or a graft copolymer). This manuf. can be used for prepg. microcapsule toners.

IT 152585-51-8P

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation) (prepn. and reaction of, microcapsule from)

RN 152585-51-8 HCAPLUS

CN Cellulose, 2-hydroxypropyl methyl ether, polymer with N-(2-aminoethyl)-1,2-ethanediamine, 2,2'-azobis[3-hydroxy-2-methylpropanenitrile], 1,3-diisocyanatomethylbenzene, methyl 2-methyl-2-propenoate, Sumidur L and N,N,N-trimethyl-2-[(2-methyl-1-oxo-2-propenyl)oxy]ethanaminium chloride (9CI) (CA INDEX NAME)

CM 1

CRN 97709-04-1 CMF Unspecified CCI MAN

CCI MAN

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

CM 2

CRN 26471-62-5 CMF C9 H6 N2 O2 CCI IDS CDES 8:ID

D1-Me

CM 3

CRN 19706-80-0 CMF C8 H12 N4 O2

CM 4

CRN 5039-78-1 CMF C9 H18 N O2 . C1

● Cl-

CM 5

CRN 111-40-0 CMF C4 H13 N3

 $H_2N-CH_2-CH_2-NH-CH_2-CH_2-NH_2$ 

CM 6

CRN 80-62-6 CMF C5 H8 O2

$$\begin{array}{c|c} ^{H_2C} & \text{O} \\ \parallel & \parallel \\ \text{Me-} \text{C-} \text{C-} \text{OMe} \end{array}$$

CM 7

CRN 9004-65-3

CMF C3 H8 O2 . x C H4 O . x Unspecified

CDES 8:GD

CM 8

CRN 9004-34-6

CMF Unspecified

CCI PMS, MAN

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

CM 9

CRN 67-56-1

CMF C H4 O

нзс-он

CM 10

CRN 57-55-6 CMF C3 H8 O2

ОН | Н3С-СН-СН2-ОН

indexing for cite #5

## LEVY 09/626,026

## => d ind 5

- L12 ANSWER 5 OF 7 HCAPLUS COPYRIGHT 2002 ACS
- IC ICM G03G009-08 ICS G03G009-087
- CC 74-4 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes)
- ST capsule toner vinyl polymer coating; amino glycidyl reaction capsule toner
- IT 172722-03-1P, Diethylenetriamine-2-ethylhexyl methacrylateglycidyl methacrylate-lauryl methacrylate-methacryloxyethyltrimethylammoni
  um chloride-methyl methacrylate-Metolose 65SH50-Sumidur L graft copolymer
  172722-04-2P, 4,4'-Azobis[4-cyanovaleric acid]-diethylenetriamine-Epiol
  A-2-ethylhexyl methacrylate-lauryl methacrylate-Metolose 65SH50-Sumidur
  L-trifluoroethyl methacrylate graft copolymer
  RL: IMF (Industrial manufacture); TEM (Technical or engineered material
  use); PREP (Preparation); USES (Uses)
  (core-shell capsule toner)

=> d ibib abs hitstr 6

L13 ANSWER 6 OF 7 HCAPLUS COPYRIGHT 2002 ACS 1993:179988 HCAPLUS ACCESSION NUMBER:

DOCUMENT NUMBER:

118:179988

TITLE:

Preparation of toners having polymer with desired

anion on their surface

INVENTOR(S):

Inaba, Yoshihiro; Kubo, Tsutomu; Takashima, Koichi Fuji Xerox Co., Ltd., Japan

PATENT ASSIGNEE(S):

Jpn. Kokai Tokkyo Koho, 11 pp.

SOURCE:

CODEN: JKXXAF

DOCUMENT TYPE:

Patent

LANGUAGE:

Japanese

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO. DATE	
				-
JP 04241361	A2	19920828	JP 1991-78657 19910114	Į
US 5385802	Α	19950131	US 1991-769250 19911001	_
PRIORITY APPLN. INFO.	:		JP 1990-268317 19901008	}
			JP 1991-78657 19910114	Į.

AB The title toners are prepd. in such a manner that a polymer, which is present on the surface of the toners, contg., as a monomer unit, a vinyl monomer having a quaternary ammonium group bearing p-toluenesulfonic acid ion and/or Me sulfate ion as its anionic component is ion-exchanged to convert the ion into other anion. The toners contg. a polymer having desired anions on their surfaces can be prepd. simply, and the toners show good charge stability under varied environmental conditions. Thus, EPT-1000 (magnetic powder) and binder resins were microencapsulated by using Sumidur L (isocyanate) and Metolose 65SH50. The obtained microcapsules were reacted with ethylene glycol dimethacrylate and subsequently with methacryloyloxyethyltrimethylammonium p-toluenesulfonate and Me methacrylate to form a graft copolymer on the surface and then ion-exchanged with Na 4-hydroxy-1-naphthalenesulfonate to give a capsule

146938-42-3 146938-42-3D, deriv. 146938-43-4 ΙT 146938-43-4D, reaction product with Fast Red A

RL: USES (Uses)

(coloring agent and binder microencapsulated with, for electrophotog. toner)

RN 146938-42-3 HCAPLUS

Cellulose, 2-hydroxypropyl methyl ether, polymer with 1,2-ethanediyl CN bis(2-methyl-2-propenoate), methyl 2-methyl-2-propenoate, Sumidur  $\hat{L}$  and N,N,N-trimethyl-2-[(2-methyl-1-oxo-2-propenyl)oxy]ethanaminium salt with 4-methylbenzenesulfonic acid (1:1), graft (9CI) (CA INDEX NAME)

CM

97709-04-1 CRN CMF. Unspecified CCI MAN

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

CM

CRN 97-90-5 CMF C10 H14 O4

CM 3

CRN 80-62-6 CMF C5 H8 O2

CM 4

CRN 40820-77-7 CMF C9 H18 N O2 . C7 H7 O3 S

CM 5

CRN 33611-56-2 CMF C9 H18 N O2

CM 6

CRN 16722-51-3 CMF C7 H7 O3 S

CM 7

CRN 9004-65-3

CMF C3 H8 O2 .  $\times$  C H4 O .  $\times$  Unspecified

CDES 8:GD

CM 8

CRN 9004-34-6

CMF Unspecified CCI PMS, MAN

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

CM 9

CRN 67-56-1 CMF C H4 O

нзс-он

CM 10

CRN 57-55-6 CMF C3 H8 O2

ОН | Н3С-СН-СН2-ОН

RN 146938-42-3 HCAPLUS

CN Cellulose, 2-hydroxypropyl methyl ether, polymer with 1,2-ethanediyl bis(2-methyl-2-propenoate), methyl 2-methyl-2-propenoate, Sumidur L and N,N,N-trimethyl-2-[(2-methyl-1-oxo-2-propenyl)oxy]ethanaminium salt with 4-methylbenzenesulfonic acid (1:1), graft (9CI) (CA INDEX NAME)

CM I

CRN 97709-04-1 CMF Unspecified CCI MAN

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

CM 2

CRN 97-90-5 CMF C10 H14 O4

CM 3

CRN 80-62-6 CMF C5 H8 O2

CM 4

CRN 40820-77-7 CMF C9 H18 N O2 . C7 H7 O3 S

CM 5

CRN 33611-56-2 CMF C9 H18 N O2

CM 6

CRN 16722-51-3 CMF C7 H7 O3 S

CM 7

CRN 9004-65-3 CMF C3 H8 O2 . x C H4 O . x Unspecified CDES 8:GD

CM 8

CRN 9004-34-6 CMF Unspecified CCI PMS, MAN

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

CM 9

CRN 67-56-1 CMF C H4 O

нзс-он

```
CM
               10
          CRN
               57-55-6
          CMF
              C3 H8 O2
    ОН
{\rm H_3C-CH-CH_2-OH}
RN
    146938-43-4 HCAPLUS
CN
    Cellulose, 2-hydroxypropyl methyl ether, polymer with ar, ar-disocyanato-
     ar-methylbenzenamine, 1,2-ethanediyl bis(2-methyl-2-propenoate), methyl
     2-methyl-2-propenoate, 2,2',2''-[1,2,3-propanetriyltris(oxymethylene)]tris
     [oxirane], Sumidur L and N, N, N-trimethyl-2-[(2-methyl-1-oxo-2-
     propenyl)oxy]ethanaminium methyl sulfate, graft (9CI) (CA INDEX NAME)
     СМ
          1
     CRN
          97709-04-1
          Unspecified
     CMF
     CCI
         MAN
*** STRUCTURE DIAGRAM IS NOT AVAILABLE ***
     CM
     CRN
          30677-40-8
     CMF
          C9 H7 N3 O2
     CCI
         IDS
     CDES 8:ID, RING
   D1-Me
  D1-NH2
2 ( D1-NCO )
     CM
          3
     CRN
         13236-02-7
     CMF C12 H20 O6
```

CM 4

CRN 97-90-5 CMF C10 H14 O4

CM 5

CRN 80-62-6 CMF C5 H8 O2

$$\begin{array}{ccc} ^{H_2C} & \text{O} \\ \parallel & \parallel \\ \text{Me-} \text{C-} \text{C-} \text{OMe} \end{array}$$

CM 6

CRN 9004-65-3

CMF C3 H8 O2 . x C H4 O . x Unspecified

CDES 8:GD

CM 7

CRN 9004-34-6

CMF Unspecified

CCI PMS, MAN

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

CM 8

CRN 67-56-1 CMF C H4 O

нзс-он

CM 9

CRN 57-55-6 CMF C3 H8 O2

ОН | Н3С-СН-СН2-ОН

CM 10

CRN 6891-44-7 CMF C9 H18 N O2 . C H3 O4 S

CM 11

CRN 33611-56-2 CMF C9 H18 N O2

CM 12

CRN 21228-90-0 CMF C H3 O4 S

Me-0-SO3-

RN 146938-43-4 HCAPLUS

CN Cellulose, 2-hydroxypropyl methyl ether, polymer with ar, ar-diisocyanato-ar-methylbenzenamine, 1,2-ethanediyl bis(2-methyl-2-propenoate), methyl 2-methyl-2-propenoate, 2,2',2''-[1,2,3-propanetriyltris(oxymethylene)]tris [oxirane], Sumidur L and N,N,N-trimethyl-2-[(2-methyl-1-oxo-2-propenyl)oxy]ethanaminium methyl sulfate, graft (9CI) (CA INDEX NAME)

CM 1

CRN 97709-04-1 CMF Unspecified CCI MAN

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

CM 2

CRN 30677-40-8 CMF C9 H7 N3 O2 CCI IDS CDES 8:ID, RING

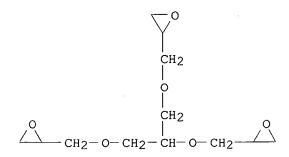
D1-Me

D1-NH<sub>2</sub>

# 2 ( D1-NCO )

CM 3

CRN 13236-02-7 CMF C12 H20 O6



CM 4

CRN 97-90-5 CMF C10 H14 O4

CM 5

CRN 80-62-6 CMF C5 H8 O2

CM 6

CRN 9004-65-3 CMF C3 H8 O2 . x C H4 O . x Unspecified CDES 8:GD

CM 7

CRN 9004-34-6 CMF Unspecified CCI PMS, MAN

\*\*\* STRUCTURE DIAGRAM IS NOT AVAILABLE \*\*\*

CM 8

CRN 67-56-1 CMF C H4 O

нзс-он

CM 9.

CRN 57-55-6 CMF C3 H8 O2

$$^{\mathrm{OH}}_{\mid}$$
 
$$_{\mathrm{H_3C-CH-CH_2-OH}}^{\mid}$$

CM 10

CRN 6891-44-7 CMF C9 H18 N O2 . C H3 O4 S

CM 11

CRN 33611-56-2 CMF C9 H18 N O2

CM 12

CRN 21228-90-0 CMF C H3 O4 S

Me-0-SO3-

#### => d ibib abs hitstr 7

L13 ANSWER 7 OF 7 HCAPLUS COPYRIGHT 2002 ACS ACCESSION NUMBER: 1992:107062 HCAPLUS

DOCUMENT NUMBER: 116:107062

Hydrophilic polyurethanes containing double bonds TITLE: Kobayashi, Tatsuhiko; Uchida, Shinji; Ota, Takayuki INVENTOR(S):

PATENT ASSIGNEE(S): Mitsubishi Kasei Corp., Japan SOURCE: Jpn. Kokai Tokkyo Koho, 8 pp.

CODEN: JKXXAF

DOCUMENT TYPE:

Patent

LANGUAGE:

Japanese

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
				<del></del>
JP 03250014	A2	19911107	JP 1990-48020	19900228
TP 2985210	B2	19991129		

AΒ The title polymers having good soly. and dispersibility in org. solvents are prepd. by copolymn. of hydrophilic vinyl monomers with double bond-contg. polyurethane obtained by reacting 0.05-1.4 mol vinyl monomer contg. 1 OH or NCO group with 1 mol polyurethane (no.-av. mol. wt. 2000-100,000) having active H group or NCO group at both ends. Polypropylene glycol (mol. wt. 2020) 1010, IPDI 88.8, and dibutyltin dioctoate 0.22 part were heated at 80.degree. for about 5~h to obtain a OH-terminated polyurethane (no.-av. mol. wt. 10,990) which was then heated with 17.0 parts 2-methacryloyloxyethyl isocyanate at 80.degree. for about 8 h to obtain polyurethane (I) contg. double bonds in the end groups with no.-av. mol. wt. 11,160. I 94, polyethylene glycol monomethacrylate (mol. wt. 440) 56, MEK 350, and AIBN 3 parts were heated at 70.degree. for about 8 h to give a 30%-solids hydrophilic polyurethane soln.

#### ΙT 139385-60-7P

RL: IMF (Industrial manufacture); PREP (Preparation) (manuf. of, hydrophilic, sol. and dispersible in org. solvents)

RN 139385-60-7 HCAPLUS

Ethanaminium, N,N,N-trimethyl-2-[(2-methyl-1-oxo-2-propenyl)oxy]-, CN chloride, polymer with .alpha.-hydro-.omega.-hydroxypoly[oxy(methyl-1,2ethanediyl)], 2-isocyanatoethyl 2-methyl-2-propenoate and 5-isocyanato-1-(isocyanatomethyl)-1,3,3-trimethylcyclohexane (9CI) (CA INDEX NAME)

CM 1

CRN 30674-80-7 CMF C7 H9 N O3

CM 2

25322-69-4 CRN CMF (C3 H6 O) n H2 O CCI IDS, PMS

CDES 8:ID

$$HO = \begin{bmatrix} (C_3H_6) - O \end{bmatrix}_n F$$

CM 3

CRN 5039-78-1 CMF C9 H18 N O2 . Cl

● c1-

CM 4

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L12 ANSWER 7 OF 7 HCAPLUS COPYRIGHT 2002 ACS IC ICM C08J009-28

ICS C08G018-83; C08J009-28

ICI C08L075-04

CC 42-10 (Coatings, Inks, and Related Products) Section cross-reference(s): 40, 43

porous sheet polyurethane coating; emulsion polyurethane coating; moisture permeability polyurethane coating; nylon fabric porous polyurethane coating; hydrophobic hydrophilic polyurethane blend coating; PTMG polyurethane coating; ethylene glycol polyurethane coating; acrylate polyurethane coating; glycerol methacrylate polyurethane coating; hydroxyethyl acrylate polyurethane coating

IT Polyamide fibers, uses and miscellaneous

RL: USES (Uses)

(fabrics, hydrophobic-hydrophilic polyurethane blend-coated, moisture-permeable)

IT Textiles

(cotton, hydrophilic-hydrophobic polyurethane blend-coated, moisture-permeable)

IT Coating materials

(emulsion, moisture-permeable, water-in-oil, hydrophilic-hydrophobic polyurethane blend, for porous substrates)

IT Paper

(release, hydrophilic-hydrophobic polyurethane blend-coated, moisture-permeable)

IT 131231-73-7P 131231-74-8P 131247-13-7P 131247-14-8P

**131247-15-9P** 131321-21-6P

RL: PREP (Preparation)

(prepn. of hydrophilic, for blending with hydrophobic polyurethanes in water-in-oil emulsions, for moisture-permeable coatings on porous substrates)

IT 61245-23-6P 94189-49-8P 131212-57-2P

RL: PREP (Preparation)

(prepn. of hydrophobic, for blending with hydrophilic polyurethanes in water-in-oil emulsions, for moisture-permeable coatings for porous substrates)

=> d ibib abs 1-24

L46 ANSWER 1 OF 24 HCAPLUS COPYRIGHT 2002 ACS ACCESSION NUMBER: 2002:107416 HCAPLUS

DOCUMENT NUMBER: 136:167821

TITLE: Polyurethane film-forming dispersions in

alcohol-water system and their manufacture for use in

RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

antimicrobial compositions

INVENTOR(S): Scholz, Matthew T.; Kantner, Steven S.; Comstock,

Kristen L.; Brown, Christopher J.

PATENT ASSIGNEE(S): 3M Innovative Properties Company, USA

SOURCE: PCT Int. Appl., 68 pp.

CODEN: PIXXD2

DOCUMENT TYPE: Patent LANGUAGE: English

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATE	ENT N	10.		KII	ND	DATE							٥.	DATE			
WO 2	20020	1024	12	 A:	 l	20020	0207				 30-U		 62	2000	1204		
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		CR,	CU,	CZ,	DE,	ĎK,	DM,	DZ,	EE,	ES,	FI,	GB,	GD,	GE,	GH,	GM,	HR,
		HU,	ID,	IL,	IN,	IS,	JP,	ΚE,	KG,	ΚP,	KR,	ΚZ,	LC,	LK,	LR,	LS,	LT,
														PL,			
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L46 ANSWER 2 OF 24 HCAPLUS COPYRIGHT 2002 ACS

ACCESSION NUMBER: 2000:807734 HCAPLUS

DOCUMENT NUMBER: 133:363129

TITLE: Antimicrobial copolymers, their production

and their use

INVENTOR(S): Ottersbach, Peter; Sosna, Friedrich

PATENT ASSIGNEE(S): Creavis Gesellschaft fuer Technologie und Innovation

m.b.H., Germany

SOURCE: Ger. Offen., 8 pp.

CODEN: GWXXBX

DOCUMENT TYPE: Patent LANGUAGE: German

FAMILY ACC. NUM. COUNT:

applicant

#### PATENT INFORMATION:

AUTHOR(S):

SOURCE:

CORPORATE SOURCE:

PATENT NO. APPLICATION NO. DATE KIND DATE A1 20001116 DE 1999-19921903 19990512 A1 20001123 WO 2000-EP2819 20000330 \_\_\_\_\_ DE 19921903 WO 2000069938 W: AU, BR, CA, CN, IL, JP, KR, NO, NZ, PL, RU, US RW: AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE 20020306 EP 1183292 A1 EP 2000-926779 20000330 R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IE, FI DE 1999-19921903 A 19990512 PRIORITY APPLN. INFO.: WO 2000-EP2819 W 20000330 AB Antimicrobial copolymers are based on 2 different aliph. unsatd. monomers having .gtoreq.1 quaternary ammonium group; a third aliph. unsatd. monomer may also be incorporated. The copolymn. may be conducted on a substrate to provide an antimicrobial coating. In an example, equal amts. of 2-(methacryloyloxy)ethyltrimethylammonium chloride and 3-(methacryloylamino)propyltrimethylammonium chloride were copolymd. to give a product effective against S. aureus and P. aeruginosa. L46 ANSWER 3 OF 24 HCAPLUS COPYRIGHT 2002 ACS ACCESSION NUMBER: 2000:431000 HCAPLUS DOCUMENT NUMBER: 133:164620 Synthesis, characterization and biocidal TITLE: properties of epoxy resins containing quaternary ammonium salts AUTHOR(S): Destais, Nadege; Ades, Dominique; Sauvet, Georges Laboratoire de Recherches sur les Macromolecules, CORPORATE SOURCE: Villetaneuse, 93430, Fr. SOURCE: Polymer Bulletin (Berlin) (2000), 44(4), 401-408 CODEN: POBUDR; ISSN: 0170-0839 PUBLISHER: Springer-Verlag DOCUMENT TYPE: Journal LANGUAGE: English AB Quaternary ammonium salts (QAS) were covalently-bound to epoxy resins of different DP in two steps: addn. of a N, N-dialkylaminoethanethiol followed by the quaternization of the tertiary amine by an alkyl bromide (C8H17Br to C14H29Br). products were characterized by 1H NMR spectroscopy. The QAS-contg. oligomers (with optional chain extender) were used as polyols to prep. polyurethane (PU) films by reaction with a triisocyanate (Tolonate HDB). The films show a good bactericidal activity against Escherichia coli, which is preserved after 6 mo of immersion in water. REFERENCE COUNT: THERE ARE 14 CITED REFERENCES AVAILABLE FOR THIS 14 RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT L46 ANSWER 4 OF 24 HCAPLUS COPYRIGHT 2002 ACS 1998:669262 HCAPLUS ACCESSION NUMBER: DOCUMENT NUMBER: 130:4791 Biocidal polyurethane and its antibacterial TITLE: properties

J. Polym. Res. (1998), 5(3), 177-186

Wang, Huei-Hsiung; Lin, Meei-Show

CODEN: JPOREP; ISSN: 1022-9760

40710, Taiwan

Grad. Inst. Text Eng., Feng Chia Univ., Taichung,

PUBLISHER: Polymer Society

DOCUMENT TYPE: Journal English LANGUAGE:

The antimicrobial finishes on the cotton fabrics has been known AB and reported recently. Particularly, the reactive-antimicrobial finishes are the most attractive. In this study, we synthesized three types of polyurethane (PU) polymers; type A (mol. wt. of polytetramethylene glycol (PTMG) is 2000), type B (mol. wt. of PTMG is 1000), and type C (mol. wt. of PTMG is 650). Firstly, the PU prepolymers were prepd. by reacting PTJG with 4,4'-diphenylmethane diisocyanate (MDI), then were extended with diethylenetriamine (DETA) (or DETA/hydrazine mixt.) to form the PU polymer. The polymer was then grafted with epichlorohydrin and further reacted with different amts. of

biocide (QAS) to form biocidal active PU quaternary ammonium salts. The biocidal

properties of the PU films were evaluated by the agar plate and the shake flask method. From the exptl. results, it demonstrates that these films and finished fabrics exhibit a high biocidal activity against Staphylococcus aureus. The biocidal activity

is found to increase with the amt. of QAS. After rinsing with water, the biocidal activity is found to increase with the amt. of QAS.

After rinsing with water, the biocidal characteristics of these films and finished fabrics remain. From IR spectra, PU

films with covalent bond of QAS show an absorption peak at 2300

cm-1, which corresponds to the presence of silicon in QAS. For the mech.

properties, the PU films with QAS sustain the mech. properties in spite of the increasing amt. of grafted QAS.

REFERENCE COUNT: 10 THERE ARE 10 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L46 ANSWER 5 OF 24 HCAPLUS COPYRIGHT 2002 ACS ACCESSION NUMBER: 1998:402515 HCAPLUS

DOCUMENT NUMBER:

129:45351

TITLE:

Antimicrobial and antithrombogenic materials

containing mucopolysaccharide quaternary

ammonium complexes

Manden, Nodriko; Seko, Masahiro; Yokota, Hideyuki; Arimori, Susumu; Tanaka, Masakazu INVENTOR(S):

PATENT ASSIGNEE(S):

SOURCE:

Toyobo Co., Ltd., Japan Jpn. Kokai Tokkyo Koho, 8 pp.

CODEN: JKXXAF

DOCUMENT TYPE:

Patent Japanese

LANGUAGE: FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

KIND DATE APPLICATION NO. DATE PATENT NO. ----------**---**----JP 10165493 A2 19980623 OTHER SOURCE(S): MARPAT 129:45351 JP 1996-330785 19961211

Title materials, useful for artificial organs and disposable medical goods, comprise (a) lipid-sol. complexes of mucopolysaccharides with trialkoxysilyl-contg. quaternary ammonium compds. and (b) org. polymers. Na heparin was treated with 3-(trimethoxysilyl)propyldimethyloctadecylammonium chloride in MeOH-H2O at 4.degree. for 15 h to give a complex. Pellethane (polyurethane) film contg. 10 phr of the complex showed retarding effect on CaCl2-induced rabbit blood plasma coaqulation and in vitro antibacterial activity against Pseudomonas aeruginosa, even after soaking into citric acid-contg. bovine blood plasma.

L46 ANSWER 6 OF 24 HCAPLUS COPYRIGHT 2002 ACS 1998:388900 HCAPLUS ACCESSION NUMBER:

DOCUMENT NUMBER: 129:58869

Antithrombogenic and antibacterial materials based on TITLE:

mucopolysaccharide quaternary

ammonium complexes and organic polymers

INVENTOR(S): Yokota, Hideyuki; Seko, Masahiro; Kadota, Noriko;

Arimori, So; Tanaka, Masakazu

Toyobo Co., Ltd., Japan PATENT ASSIGNEE(S):

SOURCE: Jpn. Kokai Tokkyo Koho, 8 pp.

CODEN: JKXXAF DOCUMENT TYPE: Patent

LANGUAGE: Japanese

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE ---------\_\_\_\_\_ A2 19980616 JP 10155898 JP 1996-314643 19961126

MARPAT 129:58869 OTHER SOURCE(S):

The materials at least contain (1) lipophilized mucopolysaccharides comprising ionic complexes of .gtoreq.1 mucopolysaccharides and quaternary ammoniums, (2) inorg. bactericides, and org. polymers. The materials show antibacterial activity against Gram-neg.

bacteria which not sufficiently susceptible to ammonium microbicides. An aq. soln. of heparin Na (I) was treated with an

aq. soln. of benzyldimethylcetylammonium chloride (II) at 4.degree. for 15

to give I-II complex. The complex was mixed with a THF soln. of Pellethane and Zeomic, and the blend was cast on a glass plate to give an

antithrombogenic **film**. The **film** inhibited growth of Pseudomonas aeruginosa. A polypropylene hollow fiber for artificial lung was coated with a THF suspension of the above blend, and tested

for biocompatibility in the femoral artery of a rabbit.

L46 ANSWER 7 OF 24 HCAPLUS COPYRIGHT 2002 ACS 1998:77062 HCAPLUS ACCESSION NUMBER:

DOCUMENT NUMBER: 128:103322

TITLE: The synthesis of biocidal PU and its

antibacterial activity

Wang, Huei-Hsiung; Iun, Dun-Iue AUTHOR(S):

Graduate School Textile Engineering Feng Chia CORPORATE SOURCE:

> University, Taichung, 40724, Taiwan Cailiao Kexue (1997), 29(4), 215-226

CODEN: TLKHAJ; ISSN: 0379-6906

Chinese Society for Materials Science PUBLISHER:

DOCUMENT TYPE: Journal LANGUAGE: Chinese

SOURCE:

AΒ We synthesized three type of polyurethane (PU) polymers; type A (mol. wt. of polytetramethylene glycol (PTMG) is 2000), type B (mol. wt. of PTMG is 1000), type C (mol. wt. of PTMG is 650). The PU prepolymers were prepd. by treating PTMG with MDI then extended with diethylenetriamine (DETA) (or DETA/hydrazine mixt.) to form the PU polymer. The polymer was grafted with epichlorohydrin and further reacted with different amts. of (triethoxysilylpropyl)dimethyloctadecylammonium chloride (QAS) biocide to form biocidal active PU quaternary ammonium salts. The biocidal properties of the PU films were evaluated by the agar plate and the shake flask method. These films and finished fabrics

exhibit a high biocidal activity against Staphylococcus aureus.

The biocidal activity is found to increase with the amt. of QAS. After rinsing with water, the biocidal activity of these films and finished fabrics still remains the biocidal characteristics. From IR spectrum, PU films with covalent bond of QAS show an absorption peak at 2300 cm-1, which is corresponding to the presence of silicon in QAS. As for the mech. properties, the PU films with QAS sustain the mech. properties in spite of the increasing amt. of grafted QAS.

L46 ANSWER 8 OF 24 HCAPLUS COPYRIGHT 2002 ACS ACCESSION NUMBER: 1995:819441 HCAPLUS

DOCUMENT NUMBER: 123:258511

TITLE: Biocidal polymers active by contact. IV.

Polyurethanes based on polysiloxanes with pendant primary alcohols and quaternary

ammonium groups

AUTHOR(S): Hazziza-Laskar, J.; Helary, G.; Sauvet, G.

CORPORATE SOURCE: Lab. Rech. Macromol., Univ. Paris-XIII, Villetaneuse,

93430, Fr.

SOURCE: J. Appl. Polym. Sci. (1995), 58(1), 77-84

CODEN: JAPNAB; ISSN: 0021-8995

DOCUMENT TYPE: Journal LANGUAGE: English

AB Functional polysiloxanes bearing both primary alcs. and quaternary ammonium salts (QAS) as lateral substituents were prepd. The synthesis involves a cohydrosilylation of allylic derivs.

(N,N-dimethylallylamine and allyloxytrimethylsilane) with various poly(di-Me siloxane-H Me siloxanes). During the quaternization of the tertiary amino groups the alc. functions are also deprotected. The hydroxyl groups allow the polysiloxane to be incorporated in polyurethane films whereas the QAS impart biocidal properties to the coating.in the case of a QAS bearing a hexadecyl substituent, a very high activity was found against Escherichia coli without any observable diffusion. The mode of action by contact between the solid polymer and the microorganisms was confirmed by the excellent durability of the biocidal power after 1 mo of immersion in water.

L46 ANSWER 9 OF 24 HCAPLUS COPYRIGHT 2002 ACS ACCESSION NUMBER: 1994:331080 HCAPLUS

DOCUMENT NUMBER: 120:331080

TITLE: Modification of central venous catheter polymers to

prevent in vitro microbial colonization

AUTHOR(S): Tebbs, S. E.; Elliott, T. S. J.

CORPORATE SOURCE: Dep. Clin. Microbiol., Queen Elizabeth Hosp.,

Birmingham, B15 2TH, UK

SOURCE: Eur. J. Clin. Microbiol. Infect. Dis. (1994), 13(2),

111-17

CODEN: EJCDEU; ISSN: 0934-9723

DOCUMENT TYPE: Journal LANGUAGE: English

AB The efficacy of an antimicrobial catheter for the prevention of bacterial colonization was investigated. The catheter was hydrophilic coated (Hydrocath) and impregnated with the quaternary ammonium antimicrobial agent, benzalkonium chloride (BZC). Microbial colonization of this central venous catheter was compared to that of polyurethane catheters with or without a hydrophilic coating. Adherence of five strains of Staphylococcus epidermidis to the three catheter types was detd. with a

microbial colonization model. Adherence of three strains of

Staphylococcus epidermidis to Hydrocath catheters was significantly reduced in comparison to polyurethane catheters.

BZC-impregnated Hydrocath catheters prevented bacterial colonization of both the internal and external catheter surfaces. These results were confirmed by SEM. The findings demonstrate that hydrophilic-coated Hydrocath catheters can inhibit bacterial adherence in vitro. Bacterial colonization was further restricted by the addn. of BZC to these coated catheters.

L46 ANSWER 10 OF 24 HCAPLUS COPYRIGHT 2002 ACS ACCESSION NUMBER: 1994:136080 HCAPLUS

DOCUMENT NUMBER: 120:136080

TITLE: Functional polysiloxanes and their preparation and use

in the manufacture of resins with permanent

biocidal properties.

INVENTOR(S): Sauvet, Georges; Helary, Gerard; Hazziza-Laskan,

Judith

PATENT ASSIGNEE(S): Groupement d'Interet Public Therapeutiques

Substitutives, Fr.

SOURCE: Fr. Demande, 17 pp.

CODEN: FRXXBL

DOCUMENT TYPE: Patent LANGUAGE: French

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
FR 2686610	A1	19930730	FR 1992-831	19920127
FR 2686610	B1	19950616		

Polysiloxanes are prepd. with quaternary ammonium side chains and reactive side groups that permit the manuf. of resins, e.g., polyesters or polyurethanes, with permanent biocidal properties. Thus, polymn. of octamethylcyclotetrasiloxane with tetramethylcyclotetrasiloxane in the presence of tetramethyldisiloxane, reaction of the product with N,N-dimethylallylamine and allyloxytrimethylsilane, and reaction of the latter product with 1-bromohexadecane in EtOH gave a product with OH and quaternary ammonium side groups. Polyurethane films prepd. from this product and Tolonate HDB exhibited good resistance to incubation of Escherichia coli.

L46 ANSWER 11 OF 24 HCAPLUS COPYRIGHT 2002 ACS

ACCESSION NUMBER: 1994:56702 HCAPLUS

DOCUMENT NUMBER: 120:56702

TITLE: Biocidal polymers active by contact. II.
Biological evaluation of polyurethane

coatings with pendant quaternary

ammonium salts

AUTHOR(S): Nurdin, N.; Helary, G.; Sauvet, G.

CORPORATE SOURCE: Lab. Recher. Macromol., Univ. Paris-XIII,

Villetaneuse, 93430, Fr.

SOURCE: J. Appl. Polym. Sci. (1993), 50(4), 663-70

CODEN: JAPNAB; ISSN: 0021-8995

DOCUMENT TYPE: Journal LANGUAGE: English

AB Films of polyurethanes were prepd. by reaction of

hydroxytelechelic polybutadienes carrying covalently bound quaternary ammonium salts with an aliph. triisocyanate.

These coatings exhibited high biocidal activity

against Gram-pos. and Gram-neg. bacteria, yeasts, and molds. Many parameters controlled the bioactivity such as the time of contact between films and bacteria, the [NCO]/[OH] ratio used to prep. the cured polyurethane, the concn. of quaternary ammonium salts in the coating, and the length of the alkyl chain from C8 to C16 linked to the quaternary N atom. A secondary phenomena of diffusion only obsd. with the shorter alkyl chains (C8 and C10) was due to synthesis residues. After these water-sol. impurities were eliminated, the biocidal activity remained excellent: then it was due only to a contact polymer bacteria.

L46 ANSWER 12 OF 24 HCAPLUS COPYRIGHT 2002 ACS 1993:605498 HCAPLUS ACCESSION NUMBER:

DOCUMENT NUMBER: 119:205498

TITLE: Biocidal polymers active by contact. III.

Aging of biocidal polyurethane

coatings in water

AUTHOR(S): Nurdin, N.; Helary, G.; Sauvet, G.

CORPORATE SOURCE: Lab. Recher. Macromol., Univ. Paris-XIII,

Villetaneuse, 93430, Fr.

J. Appl. Polym. Sci. (1993), 50(4), 671-8 CODEN: JAPNAB; ISSN: 0021-8995 SOURCE:

DOCUMENT TYPE: Journal LANGUAGE: English

Polyurethane coatings prepd. from hydroxytelechelic polybutadiene with pendant quaternary ammonium salts (QAS) were able to kill microorganisms only by contact. In order to det. the time of protection against microorganisms, these biocidal polymers were submitted to various aging conditions. The activity remained const. after exposure to a very high no. of bacteria (Escherichia coli). Nevertheless immersion in water caused a slow decrease of activity with time whatever QAS examd. The phenomenon was analyzed in 2 stages.

The first one, short (5-10 days) and limited, was due to the diffusion of a water-sol. synthesis residue. At the end of this period, the samples were still active and the activity was only due to a contact polymer-bacteria. The second stage was much lower and was accompanied by a transformation of QAS in amine. This was attributed to an equil. between QAS, amine, and alkyl bromide slowly shifted toward the formation of amine because of a weak soly. of alkyl bromide in water. Increasing the lipophilicity and bulkiness of the QAS substituents improved the durability of the biocidal activity. Some samples still

exhibited a good activity after >1 yr of aging in harsh conditions.

L46 ANSWER 13 OF 24 HCAPLUS COPYRIGHT 2002 ACS 1993:175856 HCAPLUS ACCESSION NUMBER:

DOCUMENT NUMBER: **1**18:175856

TITLE: Process for antimicrobial treatment of

polyurethane medical devices

INVENTOR(S): Laufer, Jay K.

BOC Health Care, Inc., USA PATENT ASSIGNEE(S): Eur. Pat. Appl., 11 pp. SOURCE:

CODEN: EPXXDW

DOCUMENT TYPE: Patent LANGUAGE: English

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE 19921230 EP 1992-107452 19920430 EP 520160 Α1

R: AT, BE, CH, DE, DK, ES, FR, GB, IT, LI, LU, NL, PT, SE CA 1992-2068168 19920511 CA 2068168 AA 19921229 PRIORITY APPLN. INFO.: US 1991-722784 19910628

OTHER SOURCE(S): MARPAT 118:175856

A method for impregnating a preformed polyurethane medical device with an antimicrobial quaternary

ammonium compd. (Markush structure is given) comprises the step of contacting the device with a soln. cong. said compd. in a chlorinated or fluorinated hydrocarbon solvent, and then removing the solvent. Pellethane 2363 tubing were immersed in 3% benzalkonium chloride for 1 min, then removed and dried at 65.degree. for 20 min. The impregnated tubes produced an inhibition zone of 17.0 mm diam. when placed in an agar dish which was inoculated with staphylococcus aureus.

L46 ANSWER 14 OF 24 HCAPLUS COPYRIGHT 2002 ACS

1992:209707 HCAPLUS ACCESSION NUMBER:

DOCUMENT NUMBER: 116:209707

Polymers of dihydroxy quaternary TITLE:

ammonium salts as algicidal and

microbicidal coatings

INVENTOR(S): Stovicek, Pavel

PATENT ASSIGNEE(S):

Can. SOURCE: U.S., 5 pp.

CODEN: USXXAM

DOCUMENT TYPE:

Patent English

LANGUAGE:

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

	PATENT NO.	KIND		APPLICATION NO.	DATE
PRIO AB	US 5084096 CA 1316623 RITY APPLN. INFO. The quaternary N X- and/or [R1R2N hydroxyalkyl; Y microbicides and compds. are dire polycarbonate, p microbicidal and hulls, fishing n tetradecyl chlor hydroxyphenyl)-H 1-tetradecyl(die This was reacted dissolved in BuO give a coating w	A A1 : H4 com R3CH2C = H, ( algic ctly b olyure algic ets, w ide 0. C1 0.4 thanol with H and hich c	19920128 19930420  CA pds. [4-HOC6H4CH(OH)(CH2)nOH]+ un)substituted ides. Repeatin onded to the ba thane, or alkyd idal coatings, alls, roofs, et 435, 1-diethano 35 and Al turni aminobutyl)3,3- epichlorohydrin treated with Ve ontrolled Staph	US 1990-505252 CA 1989-595836 1989-595836 Y[(CH2)nNR1R2R3]C X- [R1-R3 = (un) alkyl; X = anion] g side chains of ckbone of polyeth resins to give usable for boat c. A mixt. of he laminobutyl-3,3'- ngs 0.145 mol was bis(hydroxyphenyl	19900405 19890406 19890406 6H4OH-4]+ substituted alkyl or are prepd. as these er, polyesters,  xyl alc. 1, bis(4,4'- refluxed to give ) ammonium chloride. yether, which was . 70 hardener to S.

HCAPLUS COPYRIGHT 2002 ACS L46 ANSWER 15 OF 24

1992:201164 HCAPLUS ACCESSION NUMBER:

DOCUMENT NUMBER: 116:201164

TITLE: Antithrombogenic and antimicrobial coating compositions for medical uses

Whitbourne, Richard James; Mangan, Margaret Anne INVENTOR(S):

PATENT ASSIGNEE(S): Sterilization Technical Services, Inc., USA

SOURCE: PCT Int. Appl., 32 pp.

CODEN: PIXXD2

Patent DOCUMENT TYPE:

English LANGUAGE:

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

KIND DATE PATENT NO. APPLICATION NO. DATE ---------\_\_\_\_\_\_ A1 19920123 WO 9200747 WO 1991-US2868 19910502 W: CA RW: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LU, NL, SE CA 2087102 AA 19920113 CA 1991-2087102 19910502 A1 19941207 EP 1991-909862 19910502 EP 626854 B1 19980715 EP 626854 R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE AT 168269 E 19980815 AT 1991-909862 19910502 ES 2120961 T3 19981116 ES 1991-909862 19910502 ES 2120961 T3 19981116 ES 1991-909862 19910502 19900712 PRIORITY APPLN. INFO.: US 1990-551924 19910228 US 1991-662452 WO 1991-US2868 19910502

AΒ Antithrombogenic agents or antibiotic agents are complexed with ionic surfactants (e.g. quaternary ammonium compds.) and formulated with water-insol. polymers to provide coating materials for artificial blood vessels, catheters, etc. The invention coating compn. provides antimicrobial and/or antithrombogenic surfaces for a long time. Thus, a polyurethane tube was coated with a soln. contg. PVP 0.006, isopropanol 1.0, nitrocellulose 1.6, Et acetate 1.2, rosin ester 0.5, Bu acetate 4.8, dimethylacetamide 1.5, and Et 3-ethoxypropionate 6.1 g, then overcoated with a soln. contg. isopropanol 9.85, dimethylacetamide 1.00, and heparin-benzalkonium chloride (I) 0.15 g. The tube was dipped in a Gentian violet dye soln. and rinsed in hot running water; the tube retained the dye stain much longer than the control tube coated with I in isopropanol, without undercoat of nitrocellulose.

L46 ANSWER 16 OF 24 HCAPLUS COPYRIGHT 2002 ACS ACCESSION NUMBER: 1991:614967 HCAPLUS

DOCUMENT NUMBER: 115:214967

TITLE: Antithrombogenic and antimicrobial coating compositions for medical goods

Whitbourne, Richard James; Mangan, Margaret Anne INVENTOR(S):

PATENT ASSIGNEE(S): Sterilization Technical Services, Inc., USA

Eur. Pat. Appl., 16 pp. SOURCE:

CODEN: EPXXDW

DOCUMENT TYPE: Patent English LANGUAGE:

FAMILY ACC. NUM. COUNT: 2

PATENT INFORMATION:

PATENT NO.	KIN	D DATE	APPLICATION NO.	DATE
EP 426486	A2	19910508	EP 1990-312021	19901102
EP 426486	A3	19921125		
EP 426486	B1	19970122		
R: AT,	BE, CH,	DE, DK, ES,	FR, GB, GR, IT, LI, LU,	NL, SE
US 5069899	A	19911203	US 1989-430340	19891102
CA 2028069	AA	19910503	CA 1990-2028069	19901019
AT 147997	Ė	19970215	AT 1990-312021	19901102
ES 2099087	Т3	19970516	ES 1990-312021	19901102
PRIORITY APPLN.	INFO.:		US 1989-430340	19891102
•			US 1990-551924	19900712

Antithrombogenic, antimicrobial compns. contg. heparin and/or AΒ

antibiotics reacted with quaternary ammonium compds. or ionic surfactants and bound with water-insol. polymers are disclosed. Polyurethane tubing was coated with a soln. contg. Me Et ketone 5.0, heparin-benzalkonium chloride 0.33, isopropanol 3.7, Et 3-ethoxy propionate 0.6, Bu acetate 0.5, nitrocellulose 0.16, Et acetate 0.1, and rosin ester 0.05 g. The tubings were dried and then tested for anti-clotting properties in human plasma.

L46 ANSWER 17 OF 24 HCAPLUS COPYRIGHT 2002 ACS

ACCESSION NUMBER: 1991:601111 HCAPLUS

DOCUMENT NUMBER: 115:201111

TITLE: Microbicidal polymers containing

quaternary ammonium groups

INVENTOR(S): Hazziza, Judith; Nurdin, Nathalie; Helary, Gerard;

Sauvet, Georges

Derivery S. A., Fr. PATENT ASSIGNEE(S): Fr. Demande, 29 pp. SOURCE:

CODEN: FRXXBL

DOCUMENT TYPE:

Patent

LANGUAGE:

French

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
FR 2648676	A1	19901228	FR 1989-8344	19890622
FR 2648676	В1	19911004		

Microbicidal polymers are prepd. by grafting quaternary ammonium compds. to low-mol.-wt. vinyl polymers, by way of hydrosilylation, followed by polymn. leading to polyesters or polyurethanes. The polymers are used as adhesives, films
, or granules. 1-Dimethylaminopropyl-1,4,4'-trimethyldisiloxane (prepn. given) was reacted with hydroxy-terminal polybutadiene in H2PtCl6.H2O-contg. hexane, followed by quaternization with octyl bromide and polymn. with diphenylmethane-1,4-diisocyanate, to give a microbicidal polymer.

L46 ANSWER 18 OF 24 HCAPLUS COPYRIGHT 2002 ACS

1991:64023 HCAPLUS ACCESSION NUMBER:

DOCUMENT NUMBER: 114:64023

Microbicidal silicone rubber particles TITLE:

containing quaternary ammonium

INVENTOR(S): Yoshida, Keiji; Hamada, Mitsuo

Patent

Dow Corning Toray Silicone Co., Ltd., Japan PATENT ASSIGNEE(S):

Eur. Pat. Appl., 5 pp. SOURCE:

CODEN: EPXXDW

LANGUAGE:

English

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

DOCUMENT TYPE:

PAT	ENT NO.		KIND	DATE	APPLICATION NO. DATE
EΡ	393511		A2	19901024	EP 1990-107027 19900412
EΡ	393511		A3	19910313	
EΡ	393511		B1	19921111	
	R: BE,	DE,	FR, GB		
JP	02274763		A2	19901108	JP 1989-96849 19890417
CA	2014506		AA	19901017	CA 1990-2014506 19900412

AU 9053284 A1 19901122 AU 1990-53284 19900417

AU 625904 19920716 B2

JP 1989-96849 PRIORITY APPLN. INFO.: 19890417

The title compns., useful in paints, rubbers, and cosmetics, contain cured silicon rubber particles (av. diam. 0.5-500 .mu.m) contg. 0.1-30% silyl group-contg. quaternary ammonium salt. Thus, a liq. mixt. of a di-Me siloxane diol, Me hydrogen siloxane, octadecyldimethyl[3-(trimethoxysilyl)propyl]ammonium chloride, and Bu2Sn dioctanoate was cured at 300.degree. to give silicone rubber particles. The particles, dispersed in PhMe, were mixed with a polyurethane paint with good microbicidal activity.

L46 ANSWER 19 OF 24 HCAPLUS COPYRIGHT 2002 ACS 1990:140830 HCAPLUS ACCESSION NUMBER:

DOCUMENT NUMBER: 112:140830

TITLE: Manufacture of polyurethane-based artificial leather

Nishimura, Takeo; Nakanishi, Shinji INVENTOR(S):

PATENT ASSIGNEE(S): Kuraray Co., Ltd., Japan Jpn. Kokai Tokkyo Koho, 8 pp. SOURCE:

CODEN: JKXXAF

DOCUMENT TYPE: Patent Japanese LANGUAGE:

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

PATENT NO. KIND DATE APPLICATION NO. DATE JP 01239177 A2 19890925 JP 1988-64172 19880316 19940518

AΒ Title leather with soft touch, smooth surface, resistance to hydrolysis, weather, and cracking by bending, and having good antifungal property, is prepd. by impregnating fibrous sheets with polyurethanes contg. soft segments of av. mol. wt. 400-4000 [e.g., polytetramethylene ether, polycaprolactone, polyvalerolactone, poly(.beta.-methyl-.delta.valerolactone), poly(hexamethylene adipate)], contg. coagulation adjusters, e.g., C.gtoreq.8 alcs., alc.-modified silicones, polyoxyethylene alkylphenyl ethers, sorbitan fatty acid esters, then applied with polyurethanes contg. high alc. sulfuric acid esters, high alkyl sulfates, polyoxyethylene alkyl ethers, polyoxyethylene fatty acid esters, polyoxyethylenealkylamines, and alkyltrimethylammonium chlorides. A nonwoven fabric of 35:65 polyethylene (I)-PET blend was impregnated with a DMF soln. contg. a polyurethane (prepd. from polytetramethylene ether glycol 1200, polyethylene adipate glycol 800, MDI 1775, and ethylene glycol 378 parts) 12, stearyl alc. 1, and sorbitan tristearate 1.5%, coated at 70 g/m2 with a mixt. of the polyurethane 25, polyoxyethylene oleyl ether (II) 1, and DMF 74%, then 80 g/m2 with a mixt. of the polyurethane 16, TiO2 0.5, 2-ethylhexylsulfuric acid Na salt 1, II 0.5, water 1, and DMF 81%, coagulated with ag. DMF, dipped in hot toluene to remove I, and coated with a polyurethane finish coating to give an artificial leather having good appearance, soft touch, and crack resistance against bending.

L46 ANSWER 20 OF 24 HCAPLUS COPYRIGHT 2002 ACS 1988:530768 HCAPLUS

ACCESSION NUMBER:

DOCUMENT NUMBER: 109:130768

TITLE: Microporous waterproof coatings

INVENTOR(S): Towery, Donald R.; Hill, Berlie R.; Watson, Thomas F.;

Triplett, Benny L.

Burlington Industries, Inc., USA PATENT ASSIGNEE(S):

PCT Int. Appl., 40 pp. SOURCE:

CODEN: PIXXD2

DOCUMENT TYPE: Patent LANGUAGE: English

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

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KIND DATE
                                      APPLICATION NO. DATE
    PATENT NO.
    WO 8801570 A1 19880310 WO 1987-US2278 19870903
        W: AU, BB, BG, BR, DK, FI, HU, JP, KP, KR, LK, MC, MG, MW, NO, RO,
            SD, SU, US, US
        RW: AT, BE, BJ, CF, CG, CH, CM, DE, FR, GA, GB, IT, LU, ML, MR, NL,
            SE, SN, TD, TG
                         19910618
                                      US 1986-905135
                                                       19860909
    US 5024875
                   Α
                                     AU 1987-79687
    AU 8779687
                         19880324
                     A1
                                                       19870903
                   A1 19890712
                                      EP 1987-906267
    EP 323481
                                                       19870903
                   B1 19941207
    EP 323481
        R: AT, BE, CH, DE, FR, GB, IT, LI, LU, NL, SE
PRIORITY APPLN. INFO.:
                                    US 1986-903130
                                                       19860903
                                                    19860909
19870903
                                     US 1986-905135
                                     WO 1987-US2278
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Textiles having good water vapor permeability and waterproofing are prepd. AΒ by coating with org. polar solvent solns. of polyurethane rubbers, immersing the coated textiles in an aq. coagulating bath to ext. the solvents, washing, and drying. A compn. of Texthane 620C (urethane rubber) 47.8, Pluronic E 68 (nonionic surfactant) 3.8, 2% Carbopol 934 (acrylic resin thickener) in DMF 4.8, and DMF 43.6 parts was spaced on a polyester taffeta, washed, dried, and treated with a fluorocarbon-silicone water repellent to form a product showing moisture vapor transmission (Rm) 1533 g/m2-24 h, hydrostatic resistance (Ph) 138 Pa, and good machine laundry durability. Other examples showed coatings contg. quaternary ammonium compd. or Tinuvin 292 or mixt. of antimony oxide and hexabromocyclododecane giving good antimicrobial or UV-resistant or flame-retardant products, resp..

L46 ANSWER 21 OF 24 HCAPLUS COPYRIGHT 2002 ACS ACCESSION NUMBER: 1988:137828 HCAPLUS

DOCUMENT NUMBER:

108:137828

TITLE:

Antibiotic releasing polymers

AUTHOR(S):

Solomon, Donald D.; Sheretz, Robert J.

CORPORATE SOURCE:

Becton Dickinson Polym. Res., Dayton, OH, 45401, USA

SOURCE:

J. Controlled Release (1987), 6, 343-52

CODEN: JCREEC

DOCUMENT TYPE:

Journal

LANGUAGE:

English

AB Catheters with 2 different antibiotic coating systems were evaluated. The 1st was a simple coating of an antibiotic (dicloxacillin) complexed with a quaternary amine. The 2nd combined the latter with a polyurethane matrix. Both were

coated onto polyurethane catheter substrates for testing. The objective of such systems is to deliver prophylactic antibiotic in high concns. to the vascular access entry site and to the surrounding microenvironment of the insertion tunnel. By using a Kirby-Bauer technique and measuring inhibition zone size, it was found that both coating systems were effective in inhibiting Staphylococcus aureus. However, the half-life of the catheter antibiotic activity for the antibiotic-polymer matrix was 100% longer (24 h vs. 12 h) than the simple antibiotic-complex coated catheters. In vivo

antimicrobial inhibition. The antibiotic relasing vascular access devices can decrease the potential for vascular access site infections.

L46 ANSWER 22 OF 24 HCAPLUS COPYRIGHT 2002 ACS 1984:553876 HCAPLUS ACCESSION NUMBER:

DOCUMENT NUMBER:

101:153876

TITLE:

Polyurethane foam impregnated or coated with

fabric conditioning agent, antimicrobial

agent, and anti-discolorant

INVENTOR(S): PATENT ASSIGNEE(S): Pavlich, Mary J. Beecham, Inc., USA

SOURCE:

U.S., 5 pp.

CODEN: USXXAM

DOCUMENT TYPE:

Patent

LANGUAGE:

English

FAMILY ACC. NUM. COUNT:

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
US 4460644	Α	19840717	US 1982-453221	19821227
CA 1214302	A1	19861125	CA 1983-444117	19831222
PRIORITY APPLN. INFO.	:		US 1982-453221	19821227

AB Zn sulfate, sorbitol [50-70-4], or dextrose [50-99-7] is useful for inhibiting the discoloration of polyurethane foams impregnated with fabric conditioning agents and halide ion-contg. antimicrobial agents during storage of the foams prior to addn. to an automatic laundry dryer. Thus, a polyurethane foam was impregnated with a mixt. of Varisoft 137 47.76, Varonic 200MS (polyethylene glycol monostearate) 20.47, Zn sulfate 0.50, and an alkylbenzyldimethylammonium chloride 4.0 parts.

L46 ANSWER 23 OF 24 HCAPLUS COPYRIGHT 2002 ACS ACCESSION NUMBER: 1984:408762 HCAPLUS

DOCUMENT NUMBER:

101:8762

TITLE:

Bactericidal and antifouling properties of polymer

AUTHOR(S):

Marochko, L. G.; Kuznetsova, T. V.; Deinega, Yu. F.

CORPORATE SOURCE:

USSR

SOURCE:

Atmosferostoikie Lakokras. Pokrytiya Prognozirovanie Srokov Ikh Sluzhby, Mater. Semin. (1982), 95-8. Editor(s): Karyakina, M. I. Mosk. Dom Nauchno-Tekh.

Propag.: Moscow, USSR.

CODEN: 51NIAP Conference

DOCUMENT TYPE:

LANGUAGE:

Russian

The bactericidal and fungicidal activity of quaternary diammonium chlorides of fatty ester derivs. of ethylene-, hexamethylene-, and decamethylenediamines and Et3SnOH [994-32-1] in polyurethane , polyurethane-epoxy resin, polystyrene [9003-53-6], and polystyrene-epoxy resin latex coatings is discussed.

L46 ANSWER 24 OF 24 HCAPLUS COPYRIGHT 2002 ACS 1969:513935 HCAPLUS ACCESSION NUMBER:

DOCUMENT NUMBER:

71:113935

TITLE:

Polyurethane materials having biocidal

properties

INVENTOR(S): PATENT ASSIGNEE(S): Elmer, Otto C.; Duncan, Joe S. General Tire and Rubber Co.

SOURCE:

U.S., 5 pp. CODEN: USXXAM

DOCUMENT TYPE:

Patent

LANGUAGE:

English

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO. APPLICATION NO. DATE KIND DATE ---------19691007 US 1966-592260 US 3471423 A 19661107

AB A quaternary ammonium salt (contg. a saccharinate or cyclamate anion), which is a catalyst for the polyurethane reaction is added to a polyurethane prepolymer. The resulting cured polymer has biocidal properties and is used in millable rubber stocks, castings, sealants, adhesives, coatings, and thermoplastics. Thus, 400 parts poly(ethylene butylene adipate), 49 parts methylenebis(4-phenyl isocyanate), and 2.25 parts alkyl(dimethyl)benzylammonium saccharinate (HQ 3300) (I) were mixed to give a polyurethane gum stock which was heated 5 hrs. at 120.degree.. The stock (100 parts) was compounded with stearic acid 0.2, carbon black (FEF) 25, and 40% active dicumyl peroxide (DiCup 40-C) 4 parts and press cured at 160.degree. for 20 min. to give rubber samples with 325 psi. 100% modulus, 4050 psi. tensile strength, 520% elongation, and 64 Shore A hardness. These values were reduced on aging at 150.degree. for 200 hrs. but remained comparable with those of a similar rubber contg. no I. Both rubbers were coated with aq. dispersions of the fungi Aspergillus, Alternaria, and Penicillium and the bacteria Achrombacter and suspended over water at room temp. Samples contg. no I were heavily degraded with cracks and holes while the specimens with I contained only a few pinholes. The catalyst activity of I was better than that of laurylisoquinolinium saccharinate (II) (HQ 4330) and equiv. to ethylenediamine. Other polyurethane compns. used contained polytetramethylene glycol, 2,4-tolylene diisocyanate, and 4,4'-methylenebis(2-chloroaniline). Dimethyl(tetradecyl)benzylammonium saccharinate was also more effective than II and Me2Ph2N+Cl- as a catalyst for the reaction of BuOH with PhNCO in PhMe.